

Combustion assistée par plasma :

Du challenge de la mesure de température en combustion turbulente à un diagnostic pertinent pour les décharges nanosecondes

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GDR EMILI – 25 octobre 2023 - Nancy

REMERCIEMENTS

Une illustration puisant dans de nombreuses collaborations

G. Cléon, A. Lo, P. Vervisch, H. Ajrouche, F. Guichard, B. Lecordier, L. Lacour, D. Honoré, P. Tardiveau, A. Brisset



EXFIDIS, FAMAC, PASTEC



BIOENGINE project, co-financed by the European Union with the European regional development fund (ERDF) and by the Normandy Regional Council



RÉGION
NORMANDIE



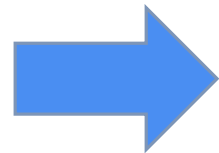
Normandie Université

INTRODUCTION

Illustrer l'intérêt du croisement des disciplines

Une opportunité : la combustion assistée par plasma

Un diagnostic d'une discipline à l'autre



La Diffusion Raman
Spontanée

COMBUSTION ET TRANSITION ENERGETIQUE

Les enjeux : une combustion neutre en carbone, efficace, propre et sûr



Émission CO₂, émission Nox, stabilité des flammes, allumage

Combustion encore 84% de la production d'énergie : la transition est une phase clé

Combustion neutre en carbone = combustion « défossilisée »,
orientée vers de nouveaux vecteurs énergétiques issues des EnR, capturer le CO₂

Focus : sur les mobilités lourdes, le transport aérien, les industries fortes émettrices en CO₂

COMBUSTION ASSISTÉE PAR PLASMA

Flame stabilization

- Steady phenomenon
- Specificity: low energy addition
 - Few % of flame power

Ignition

- Single event
- Enlarge the flammability domain
 - highly diluted reactants,
 - high-pressure conditions,
 - re-ignition under various thermodynamic conditions

- **combined production of**
 - Reactive species, radicals
 - Heat with non-equilibrium effect



**affect combustion
chemical kinetics**

CARACTÉRISER LA COMBUSTION TURBULENTE

What do we want to characterize?

- High temperature reactive flows
- Chemical reaction
 - Burning velocity...
- Mass transfer, mixing
- Heat transfer

What do we want to measure?

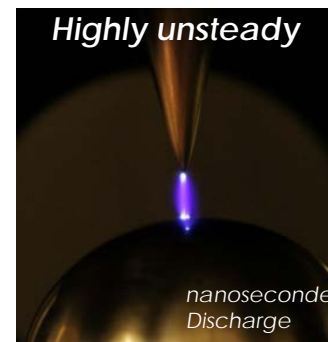
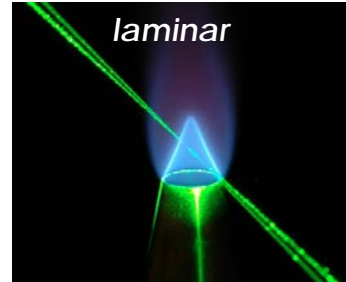
- Local composition
- Temperature
- Aerodynamics

gradients

Simultaneously

- Joined pdf
- Conditioned data

Temporal monitoring



What difficulties?

Gas

High temperature

Luminous medium

Reactive flows:
Numerous species

Small time scale:
Time resolution $< 1\mu\text{s}$

Sharp gradients:
Spatial resolution $\sim 100\mu\text{m}$

3D phenomenon

Numerous interactions

COMBUSTION TURBULENTE ET MESURE DE TEMPÉRATURE ?

Flamme : milieu hostile mais fragile

Flamme turbulente : milieu hétérogène, instationnaire (même si stationnaire en moyenne souvent)

milieu hétérogène : fort gradient, résolution spatiale $\sim 100\mu\text{m}$, résolution temporelle $< 1\mu\text{s}$

mesure locale -> Absorption

mesure instantanée ($< 1\mu\text{s}$) -> HF où beaucoup d'hypothèses sur incertitude en fonction de T, sur l'influence du quenching

ATTRAIT DE LA DIFFUSION RAMAN SPONTANÉE

😊 Diffusion de la lumière

- Quelque soit la longueur d'onde du laser
mais plus efficace avec les courtes longueurs d'onde

😊 Toutes les espèces présentes dans le volume de mesure diffusent la lumière : mesures multi-espèces

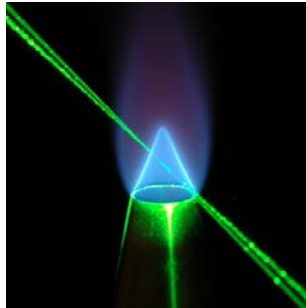
😞 Un rayonnement très faible limitant ses champs d'application depuis sa découverte

section efficace caractéristique 10^{-30} cm²/sr

à comparer à l'absorption/LIF : 10^{-18} cm²/sr

DRS : UN CHALLENGE EN COMBUSTION TURBULENTE

in flames



Gas
Low density
High temperature
Lower density
Reacting flow
Numerous species
Parasitic flame emission
LIF interferences

In turbulent flames



Short time scales
Time resolution $< 1\mu\text{s}$
Sharp gradients
Spatial resolution $\sim 100\mu\text{m}$
Linewise measurements

Diffusion Raman Scattering

Energetic pulsed laser
w/o optical breakdown
Window damage
Nonlinear phenomena

1D: imaging spectrograph

High throughput of the collection system

High efficiency of the detector

Fast gating

MESURE DE T PAR DIFFUSION RAMAN SPONTANÉE

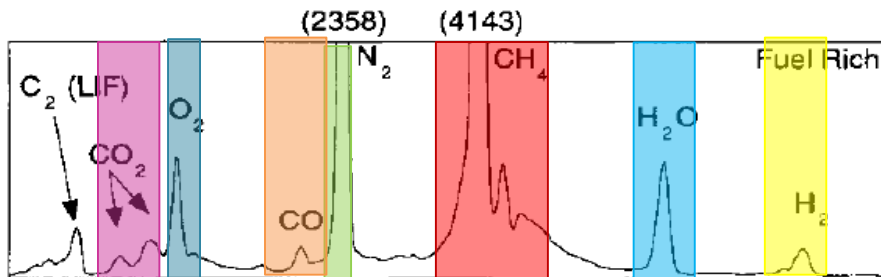
Dès 1972, proposition de mesurer T par ajustement de spectre

M. Lapp, et al, Science, 175 (1972) 1112

- o Mais trop peu de signal pour les flammes turbulentes :

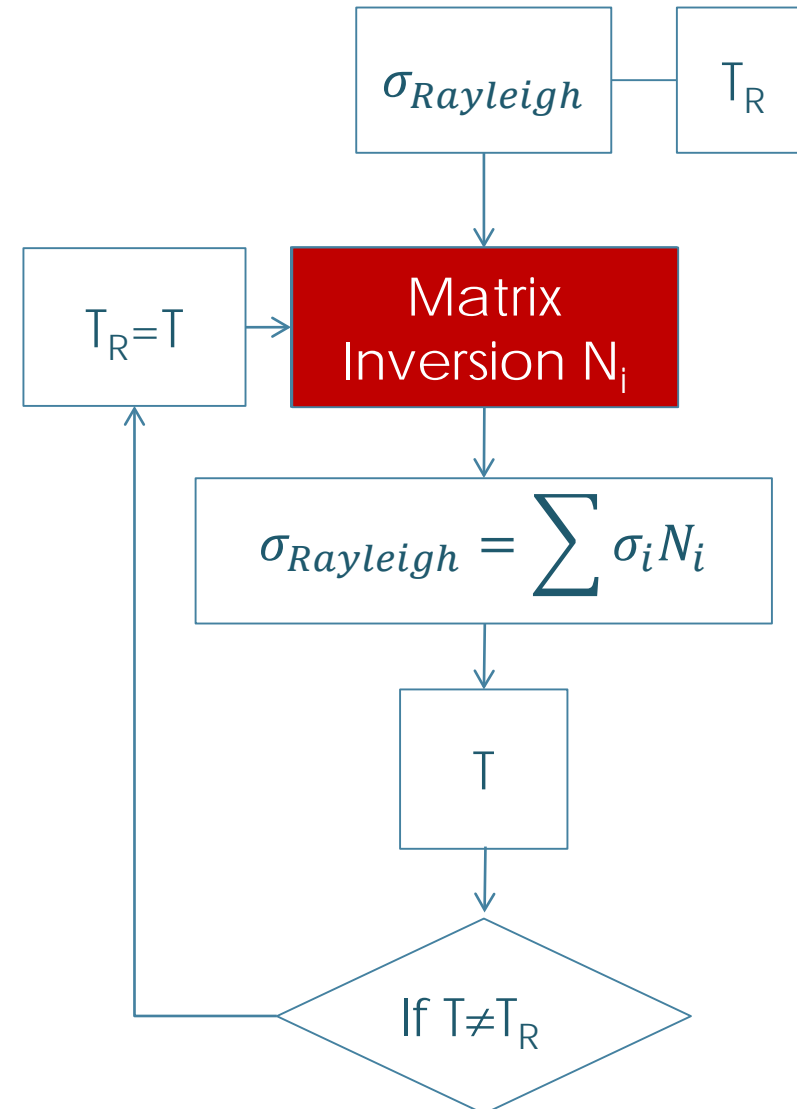
- o temps exposition court,
- o faible volumen de mesure

- o En 1988, une proposition sans ajustement de spectres : Rayleigh-Raman

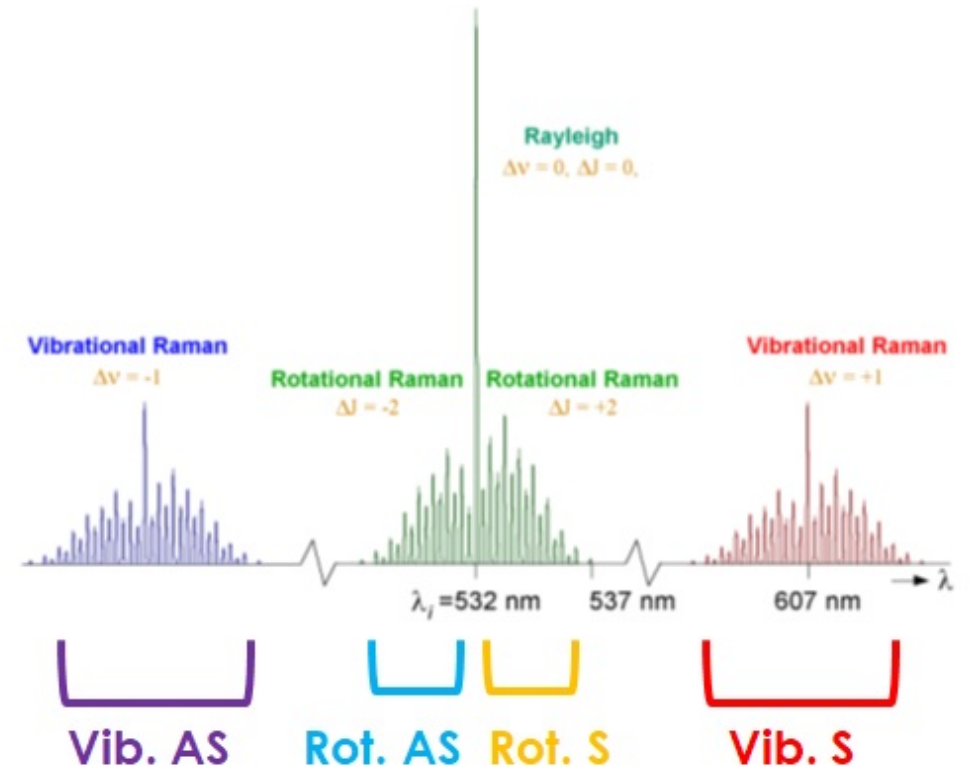
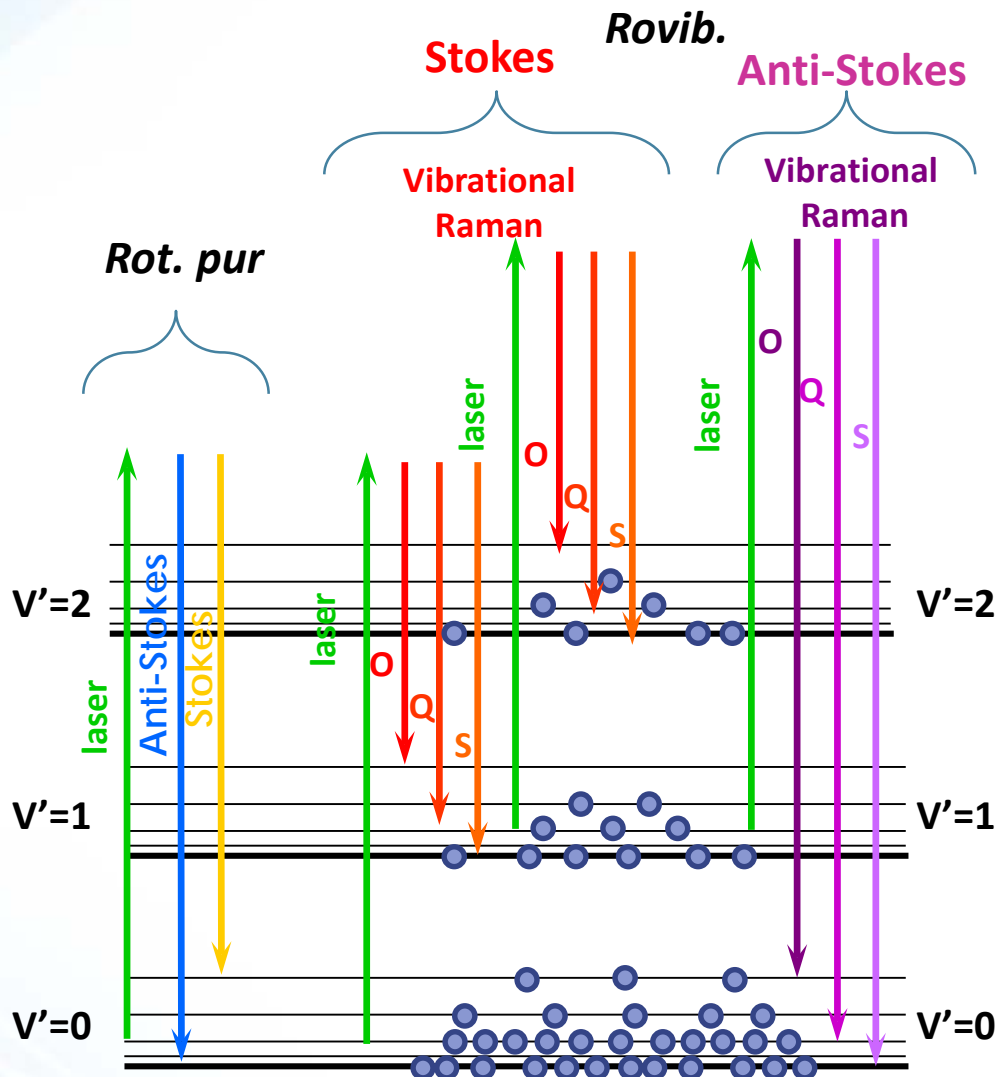


UMR 6614

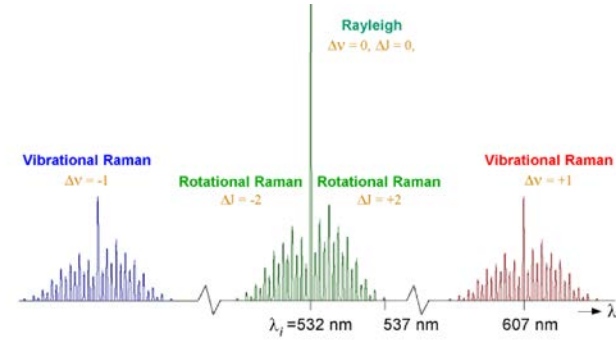
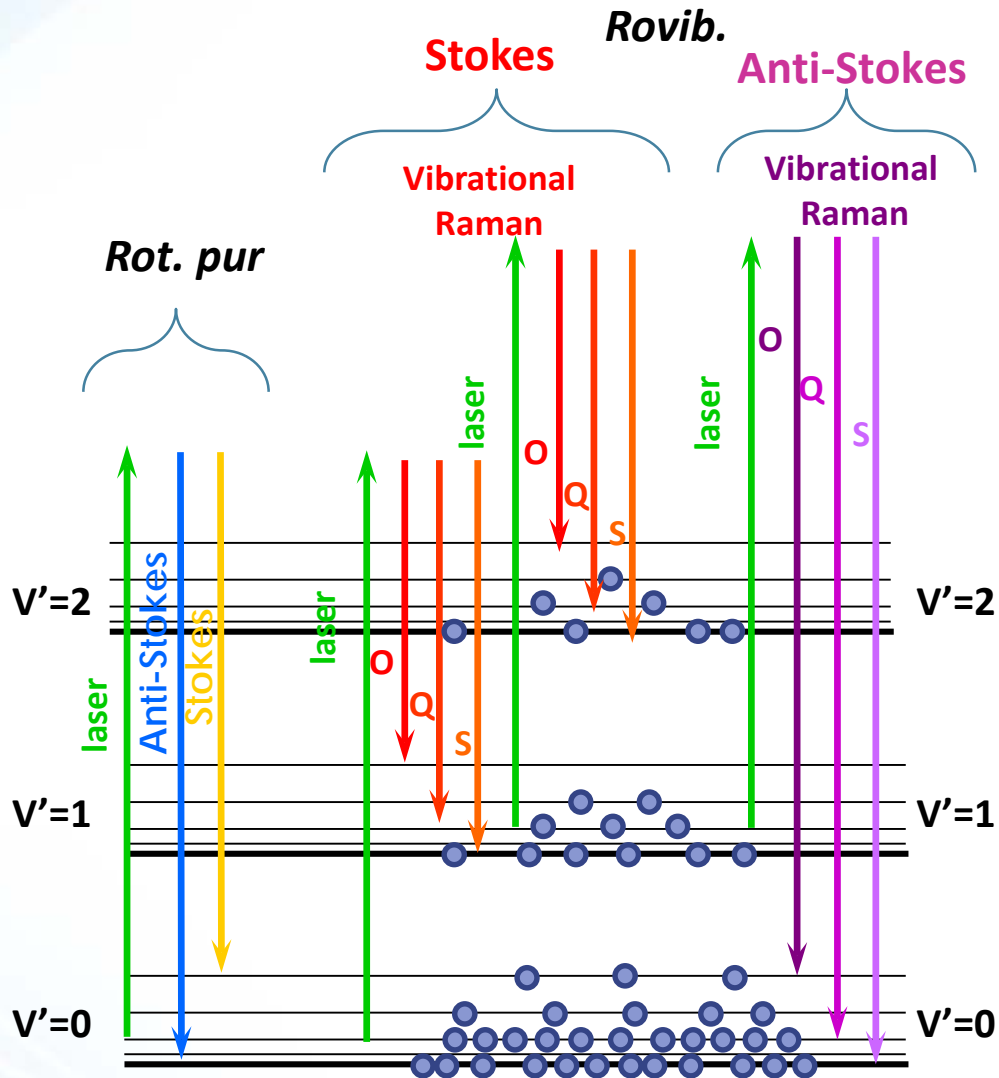
Dibble 1988, Barlow 1989



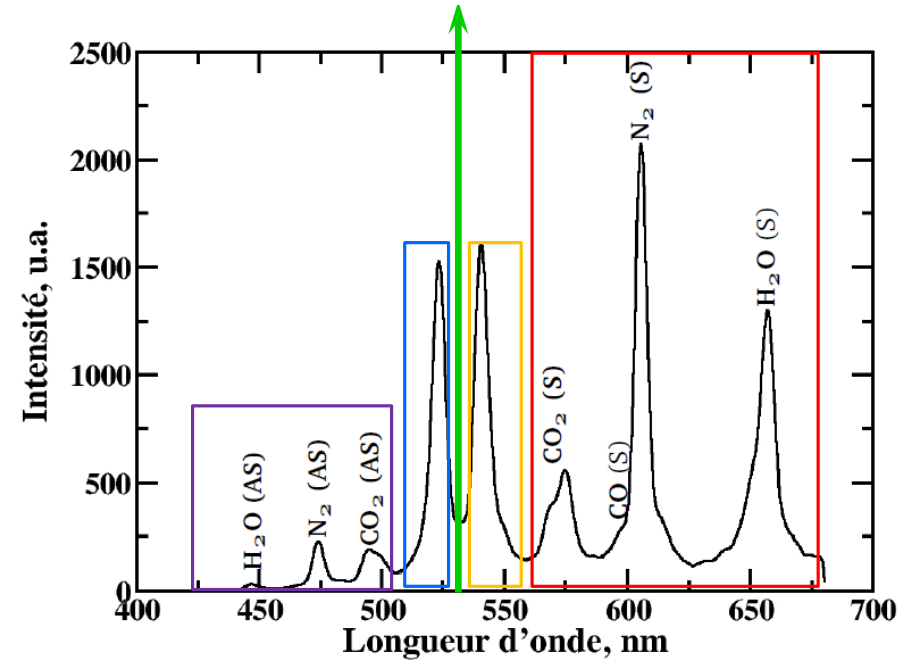
PRINCIPE DE LA DIFFUSION RAMAN : BREF RAPPEL



PRINCIPE DE LA DIFFUSION RAMAN : BREF RAPPEL

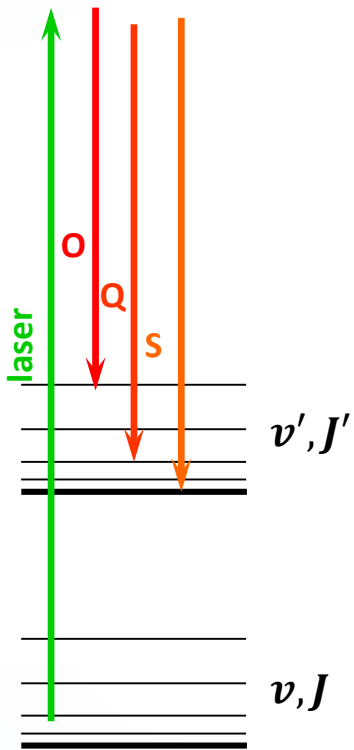


laser excitation @532 nm



AJUSTEMENT SPECTRAL

excitation



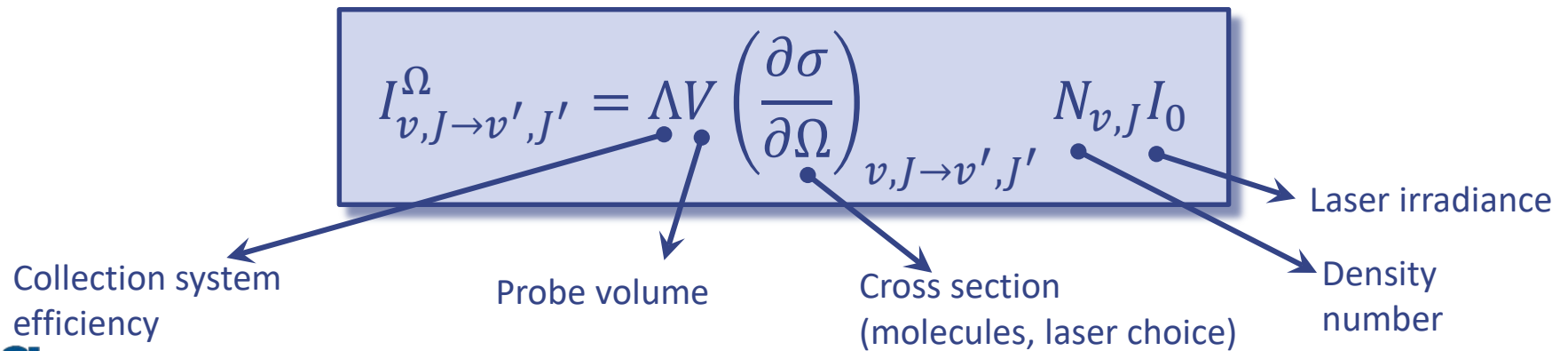
- Transition moment from (v, J) to (v', J')

$$\vec{P} = \varepsilon_0 \langle v, J | \vec{p} | v', J' \rangle = \varepsilon_0 \langle v, J | \alpha | v', J' \rangle \vec{E}$$

- Cross section

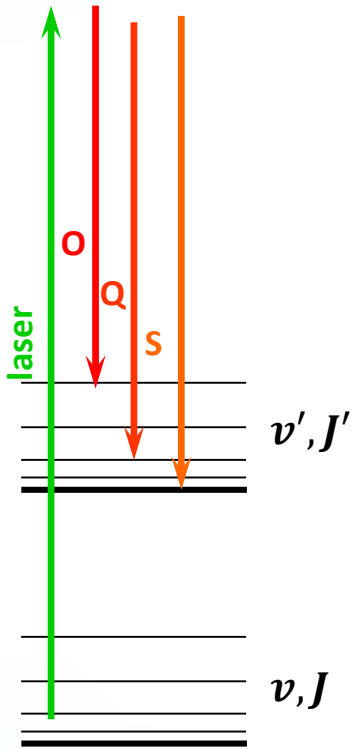
$$\left(\frac{\partial \sigma}{\partial \Omega} \right)_{v, J \rightarrow v', J'} = \pi^2 (\omega_0 - \Delta \omega_{v, J \rightarrow v', J'})^4 \langle v, J | \alpha | v', J' \rangle^2$$

- $[\alpha] = [\alpha_0] + \left(\frac{\partial [\alpha]}{\partial Q} \right)_0 Q$

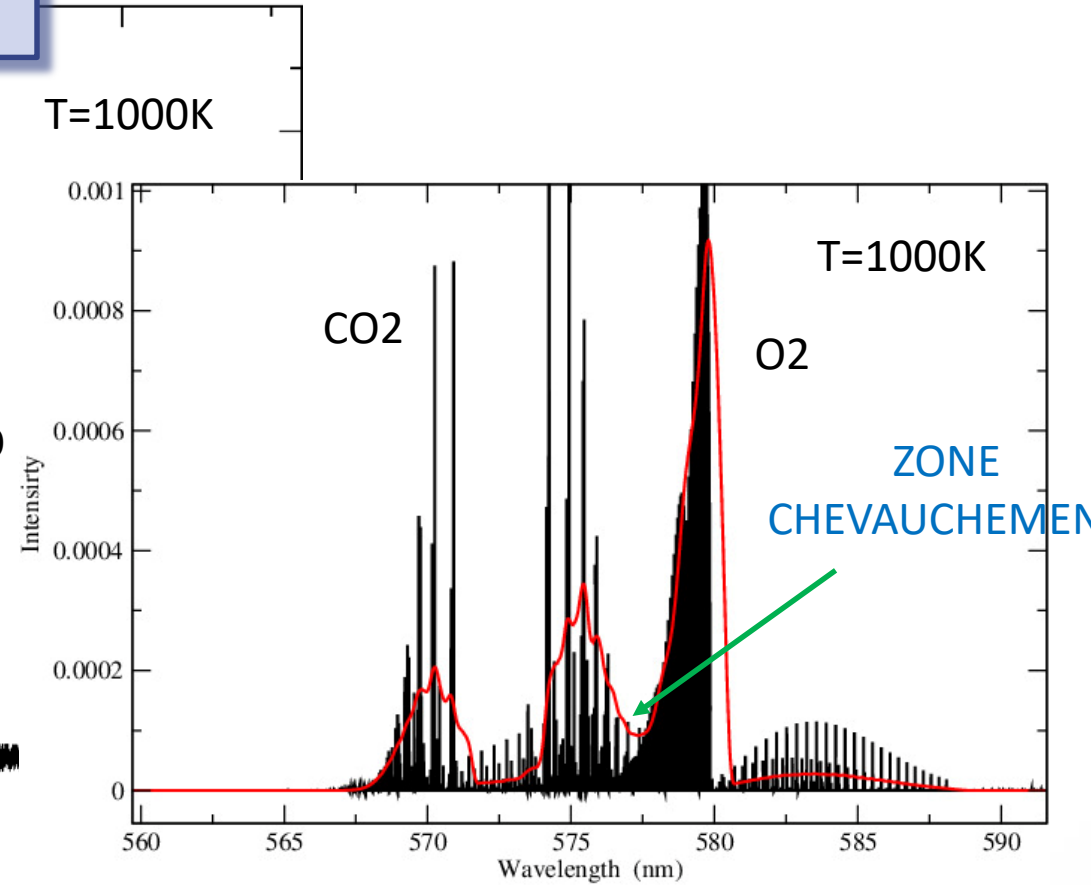
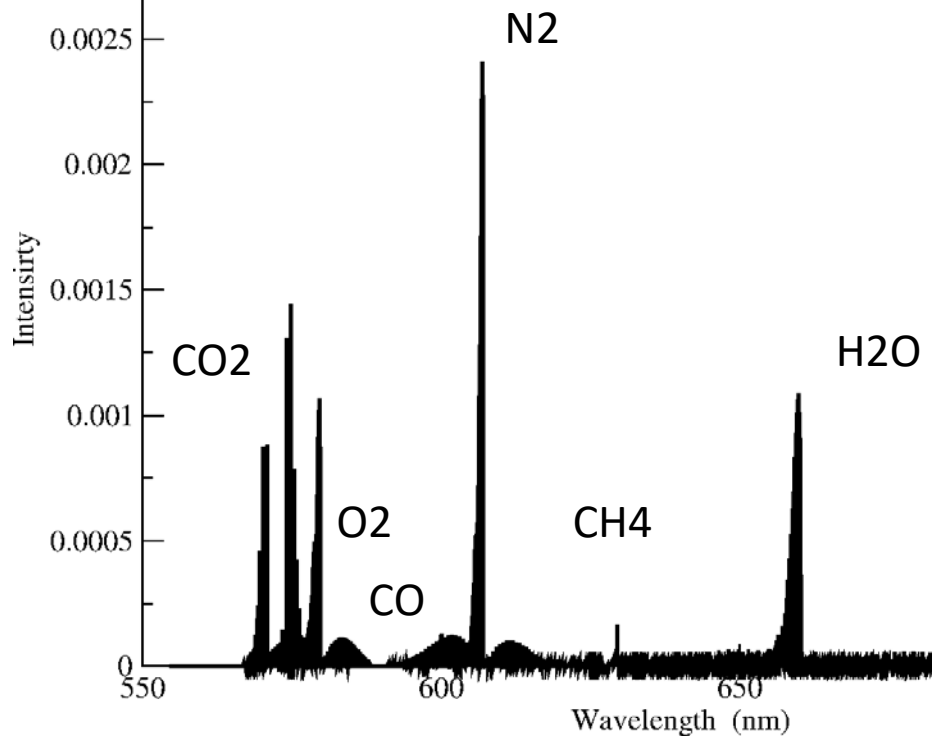


AJUSTEMENT SPECTRAL

excitation

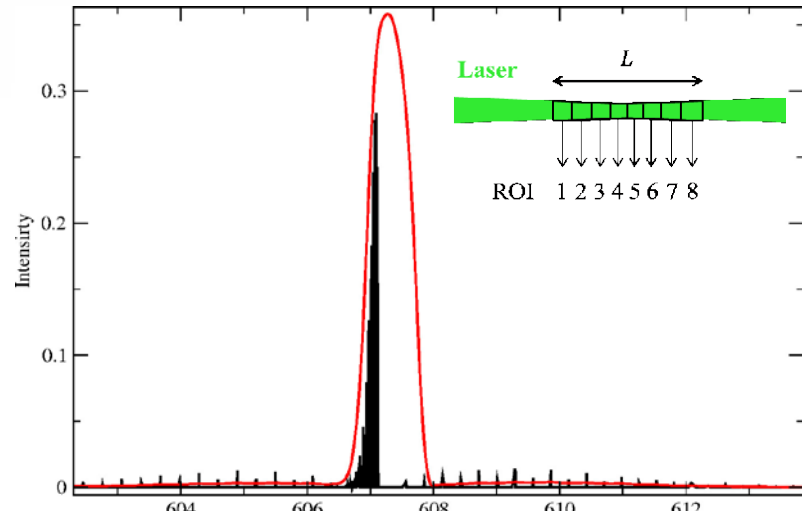


$$I_{v,J \rightarrow v',J'}^{\Omega} = \Lambda V \left(\frac{\partial \sigma}{\partial \Omega} \right)_{v,J \rightarrow v',J'} N_{v,J} I_0$$

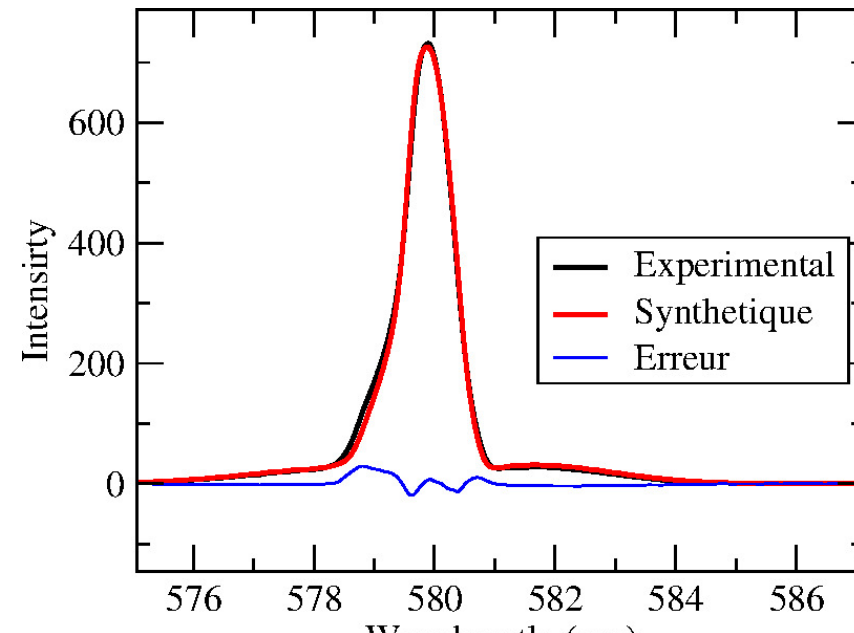
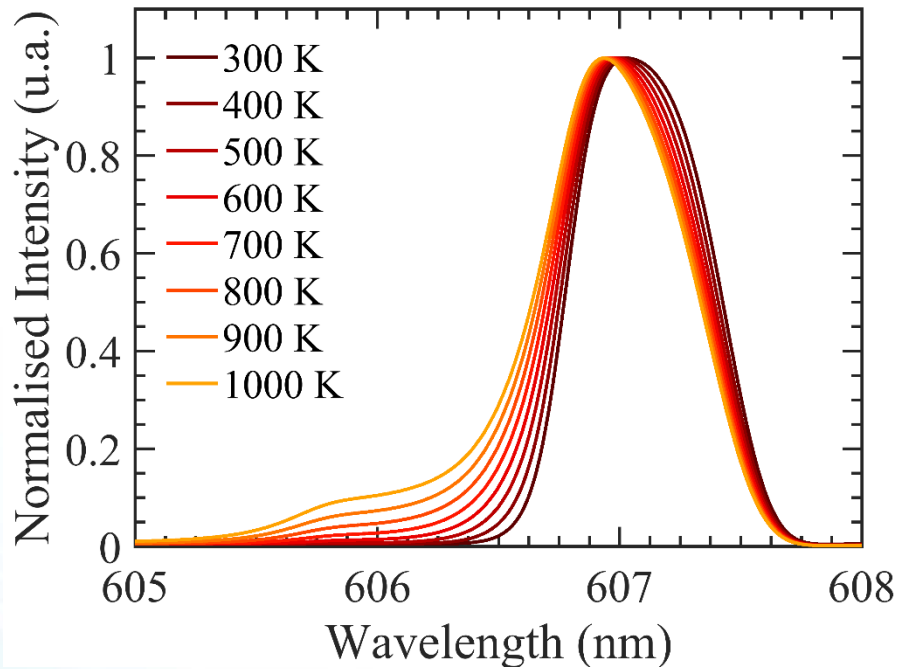
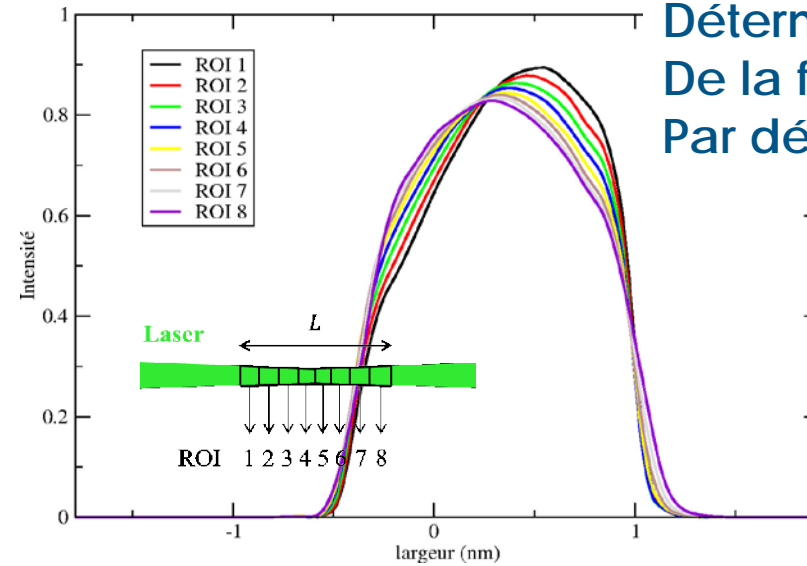


TEMPÉRATURE MALGRÉ UNE FAIBLE RÉOLUTION SPECTRALE

Spectre synthétique (N2 à 300K)

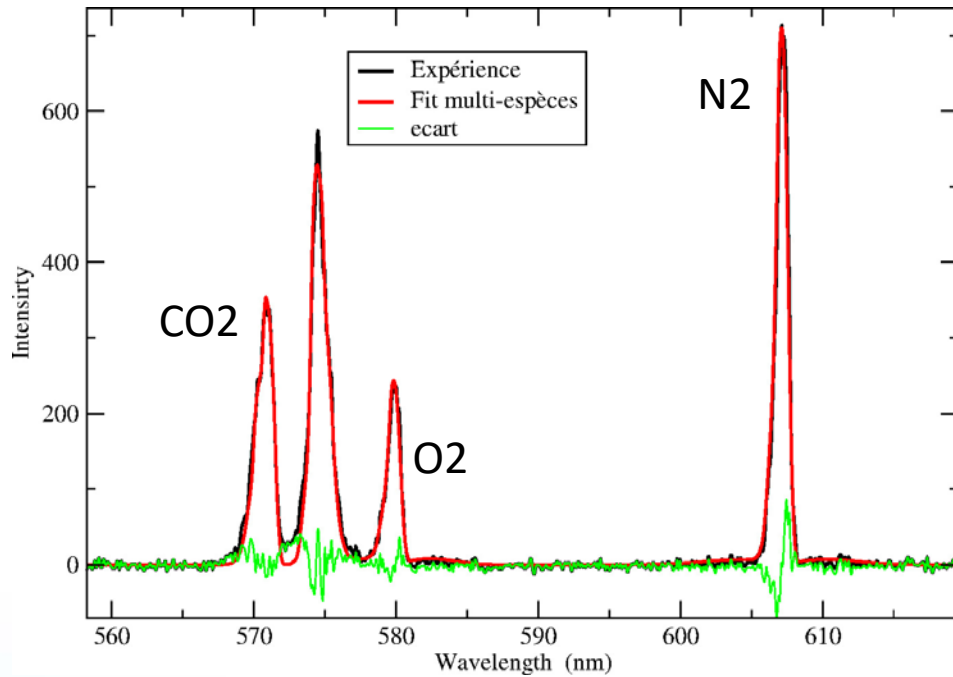


Détermination locale
De la fonction d'appareil
Par déconvolution

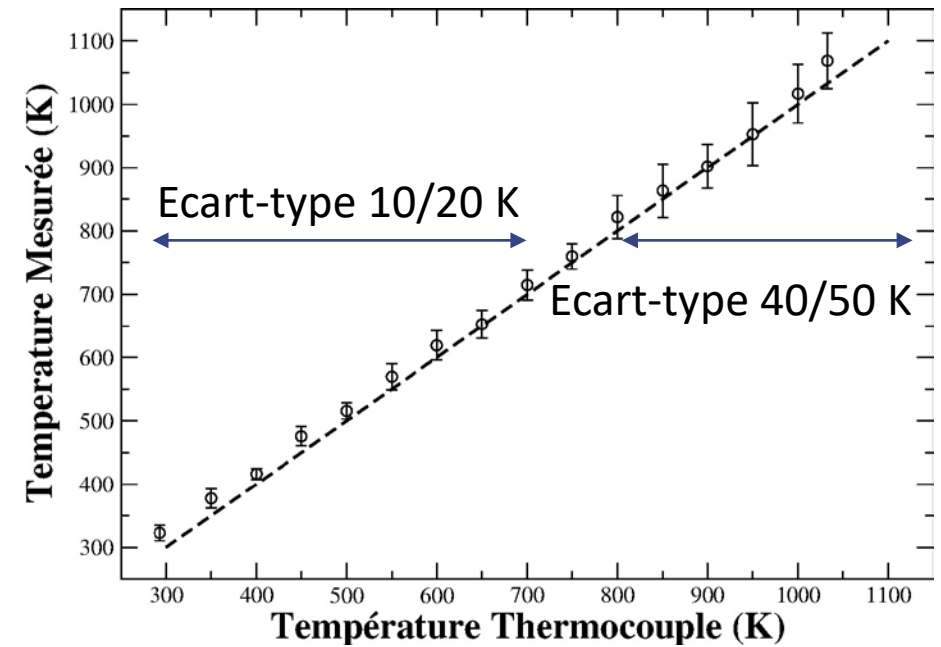


MESURE A BASSE TEMPÉRATURE

Spectre O₂/N₂/CO₂
Validation sur Jet d'Air (70%)/CO₂ (30%) chauffé de 300 à 1033K

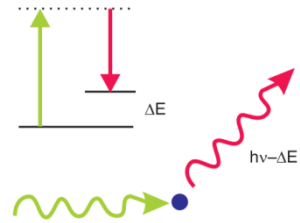


Nouvelle approche pour la T°
(Minimisation N₂/O₂/CO₂)

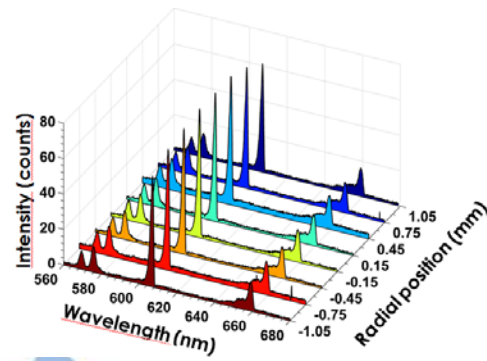
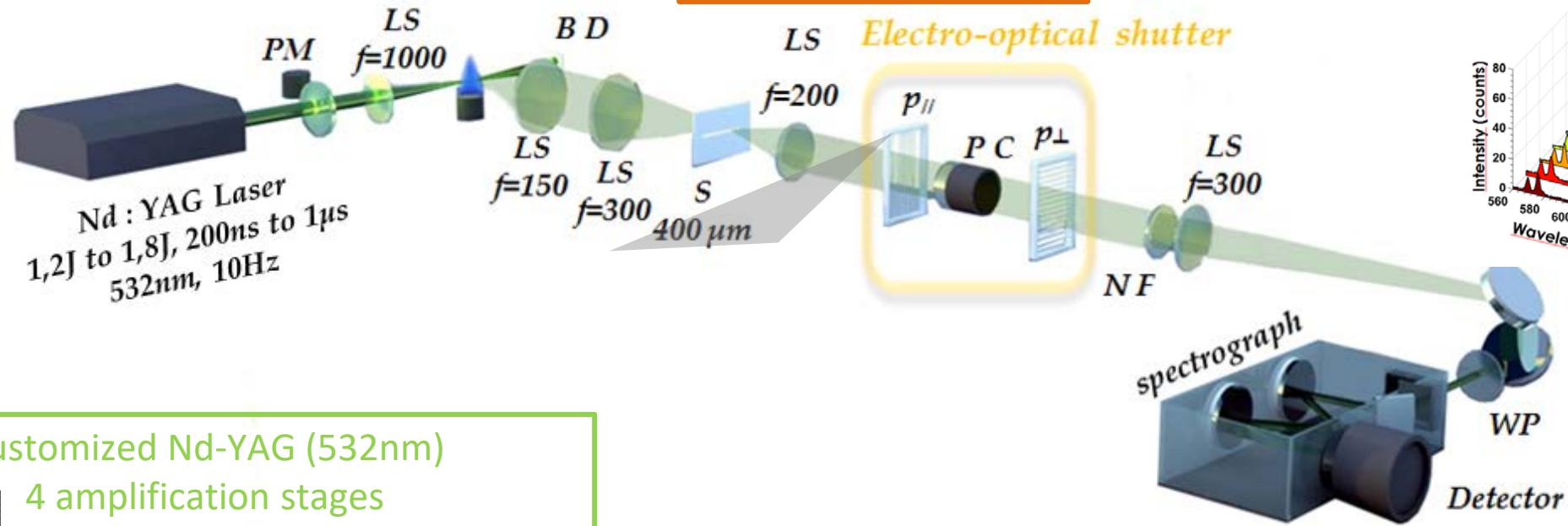


DISPOSITIF EXPÉRIMENTAL

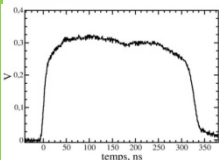
Need **high energy pulse**,
sensitive detection and
fast gating



Pockels cell +
2 crossed polarizers
1μs gate width



Customized Nd-YAG (532nm)
4 amplification stages
2 doubling crystals
1,8J for 1μs at 10 Hz



Achromat lenses of large aperture
Efficient imaging spectrograph
Ultra-sensitive BI-CCD

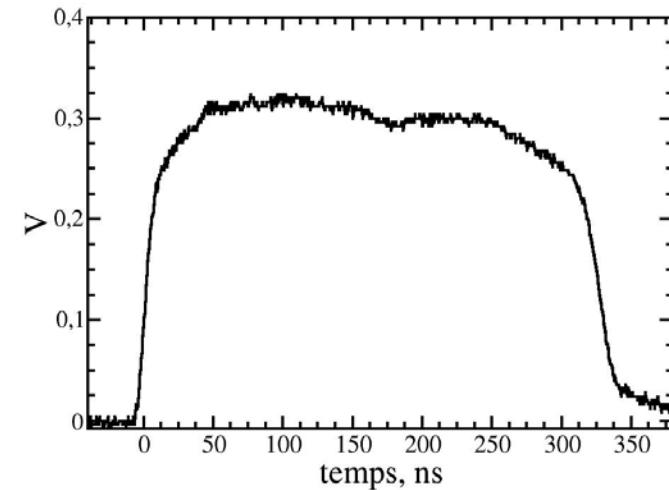
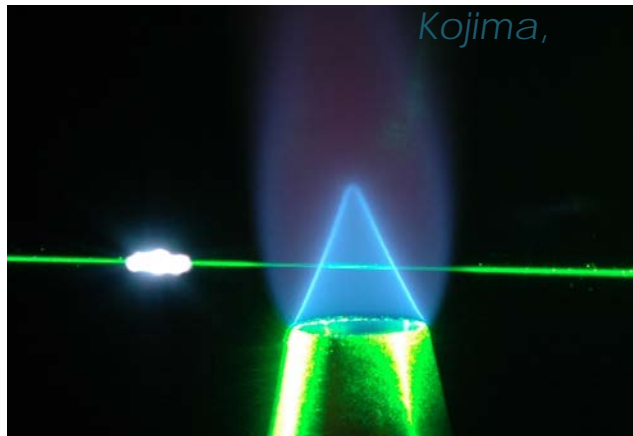
LE CHOIX DU LASER

- Minimum energy required for single-shot measurement :1 Joule (Meier, 2002)
 - Window damage
 - Non-linear effects
 - *Optical breakdown, stimulated Raman scattering*
 - Laser pulse stretching
 - *Maximal irradiance: 35 GW/cm² (Cléon, 2007)*

Dye laser , 2 μ s, 1J x 2 (mirror) *Dibble et al.(1990)*

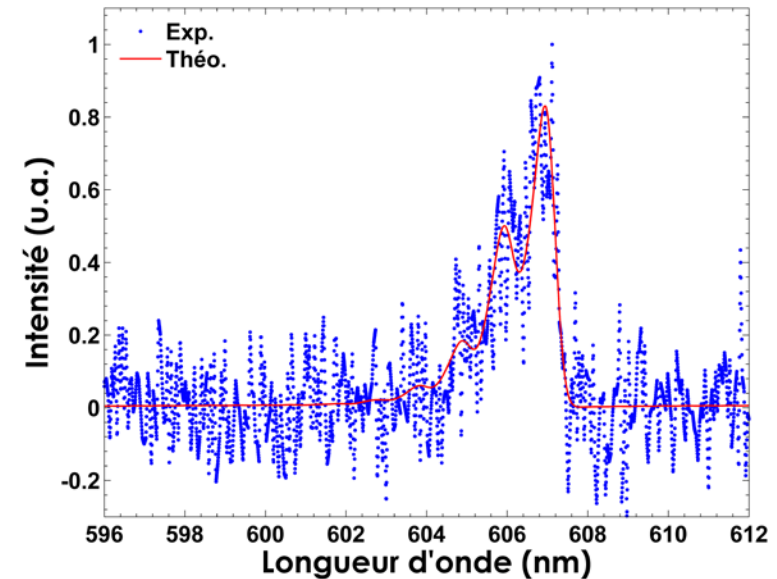
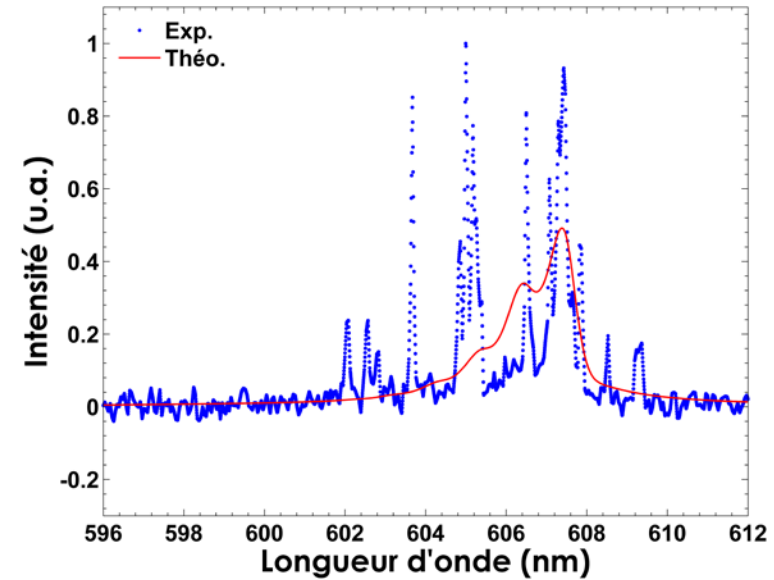
Pulse stretcher, 400 ns, 1.6J *Magnotti et al (2014)*

Long pulse laser, 300 ns-1 μ s, 1.8J *Guichard et al. (2018),*



LE CHOIX DU DETECTEUR

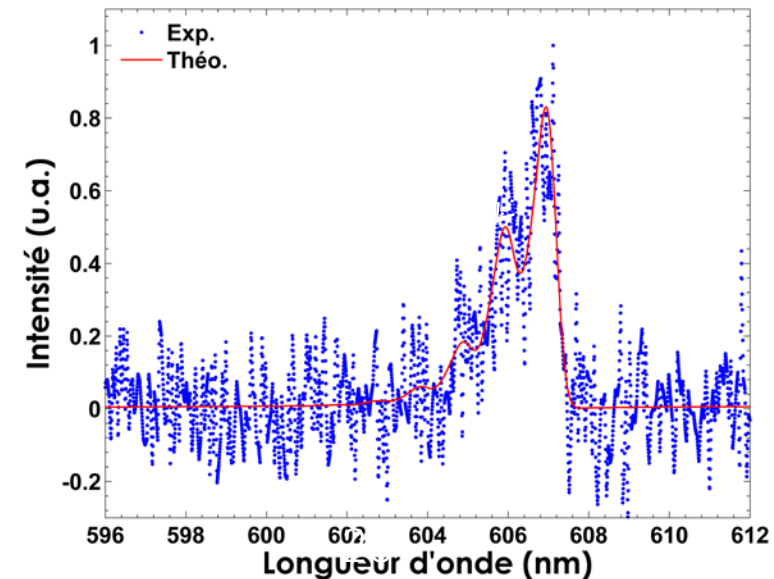
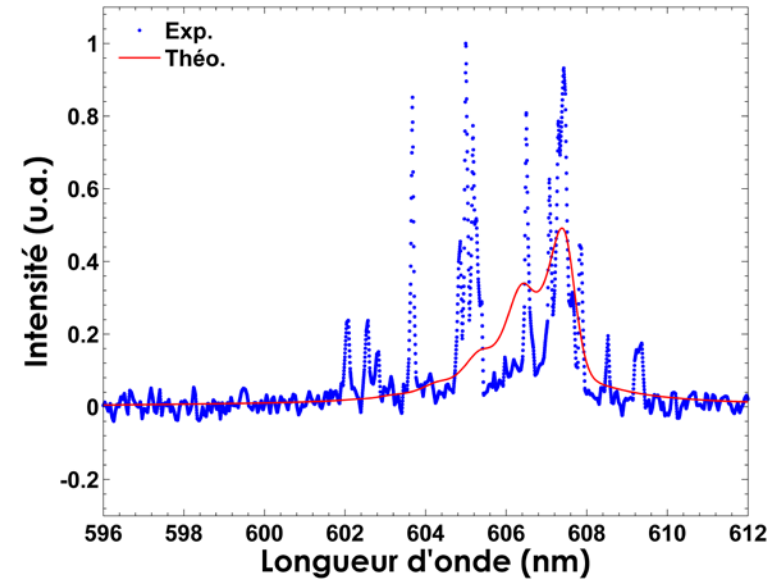
- **ICCD camera**
 - Fast gating (~1ns)
 - High shot noise
- **Back-illuminated CCD camera**
 - Very high quantum efficiency (QE>80%)
 - Low noise
 - Very high detectability
 - But full-frame architecture



Instantaneous measurement in flames: detector choice

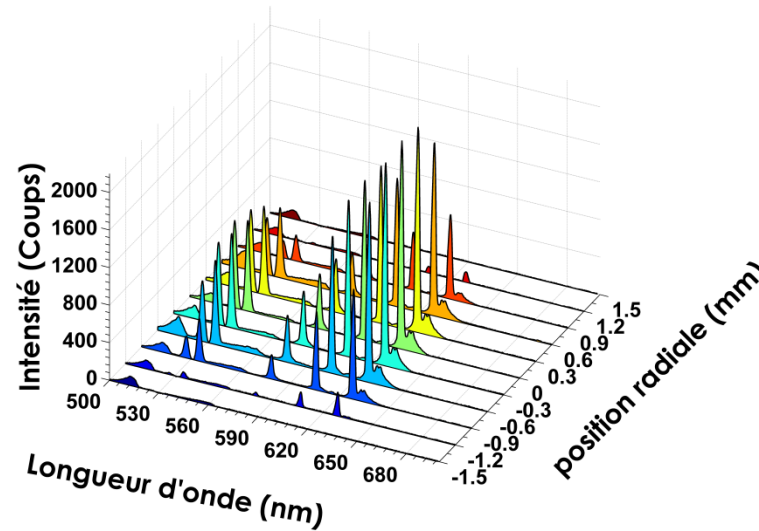
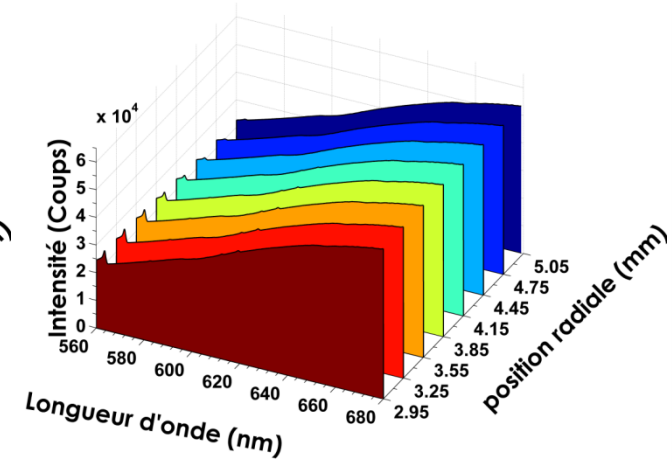
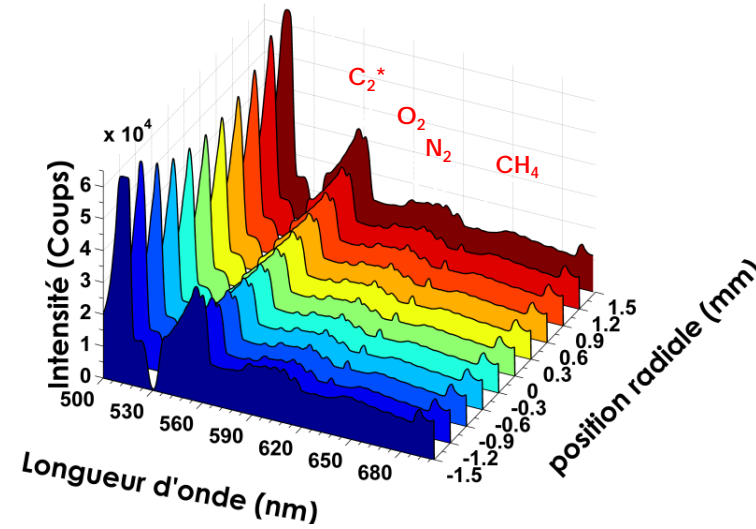
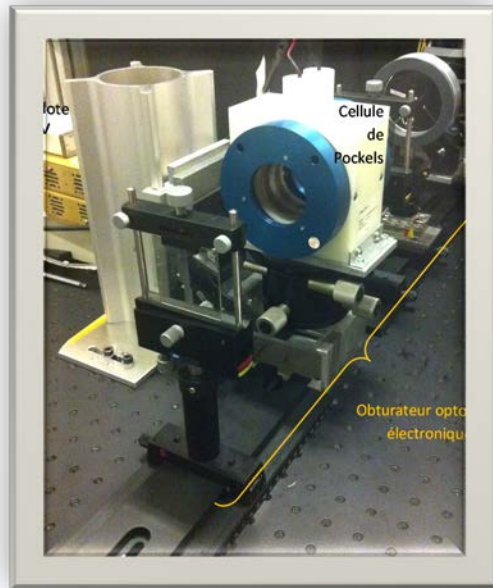
- ICCD camera
 - Fast gating (~1ns)
 - High shot noise
- Back-illuminated CCD camera
 - Very high quantum efficiency (QE>80%)
 - Low noise
 - Very high detectability
 - But full-frame architecture

➔ Impose fast shutter



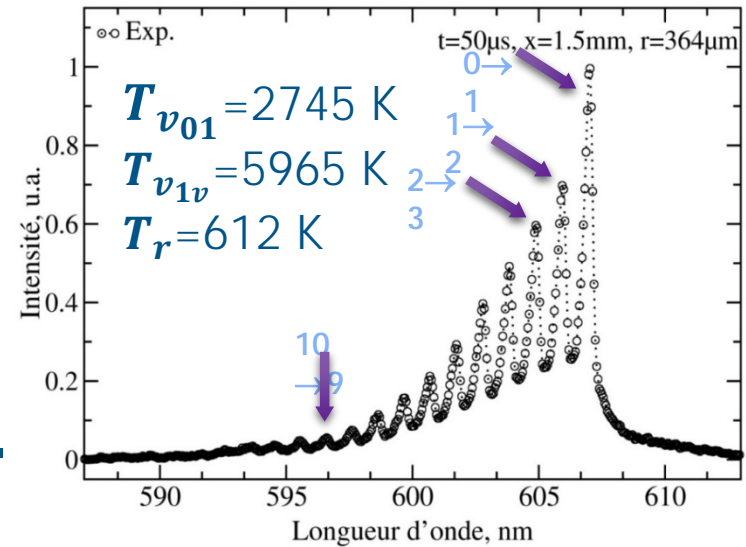
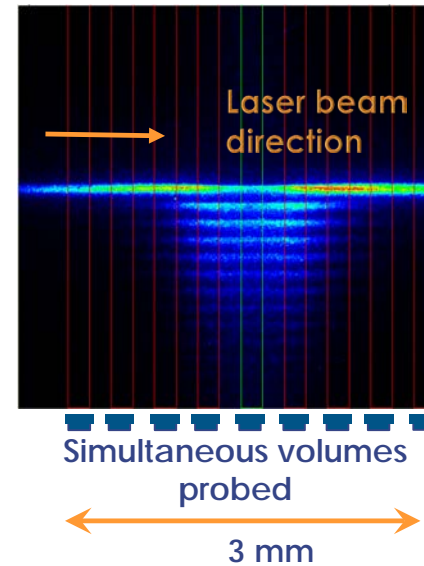
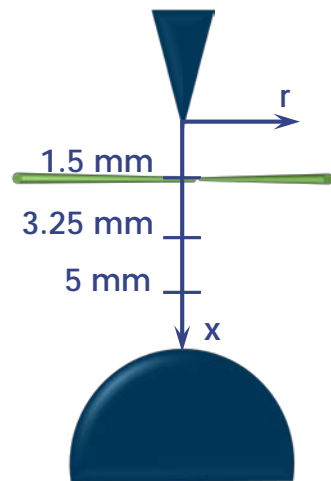
Obturation pour les caméras rétro-éclairées

- Obturateur opto-électronique
 - Cellule de Pockels entre deux polariseurs croisés



Décharge nanoseconde

INTEREST FOR INVESTIGATION OF AIR DISCHARGE



Lo et al.
Appl. Phys B(2012)

The interest of SRS is to provide :

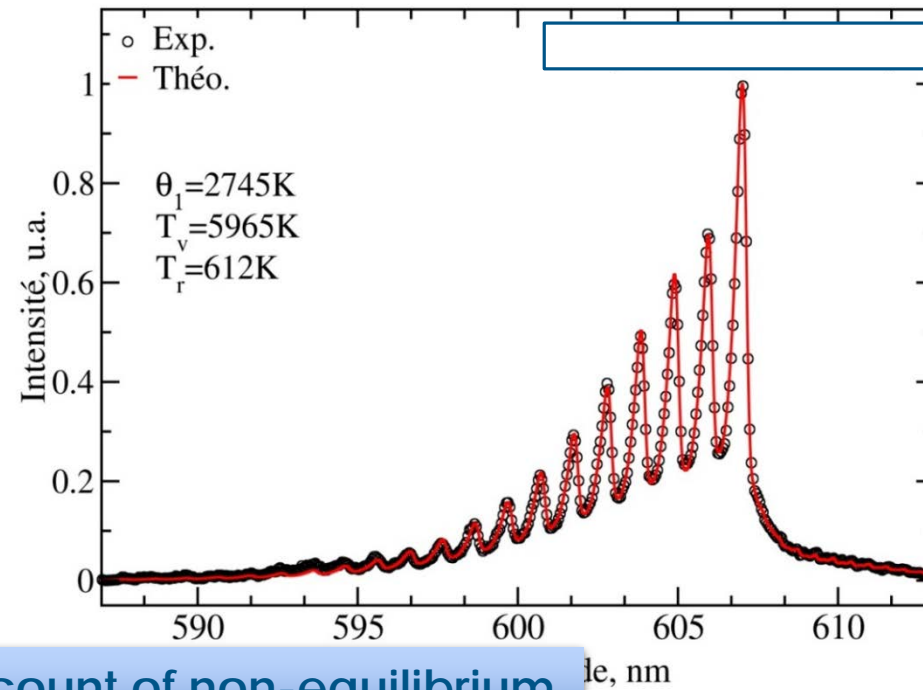
Linewise measurements : 1D profiles without scanning

Good spatial resolution : $<200 \mu\text{m}$ along r

To probe many vibrational levels simultaneously

Measurement of the vib. population up to $v=16$

INTEREST FOR INVESTIGATION OF AIR DISCHARGE



Lo et al.
Appl. Phys B(2012)

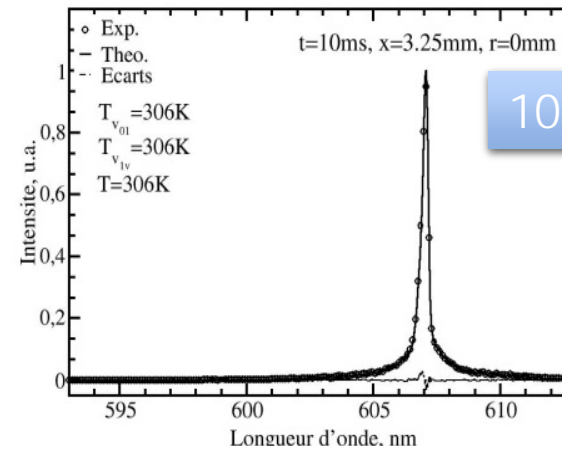
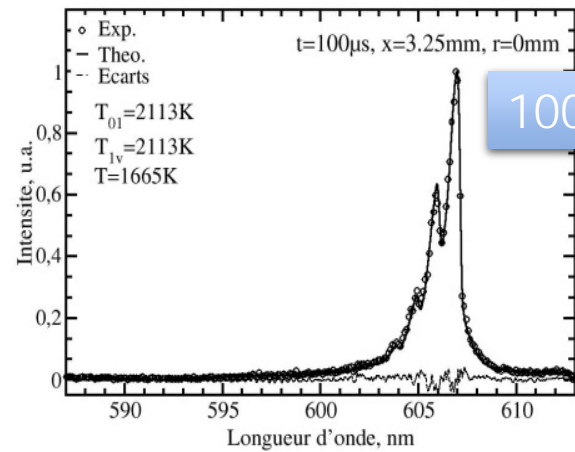
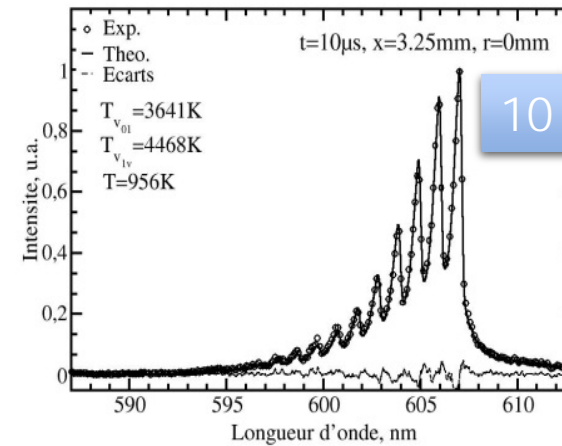
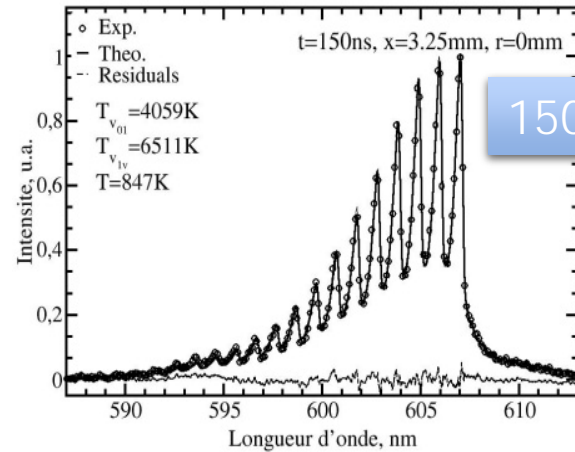
Taking into account of non-equilibrium

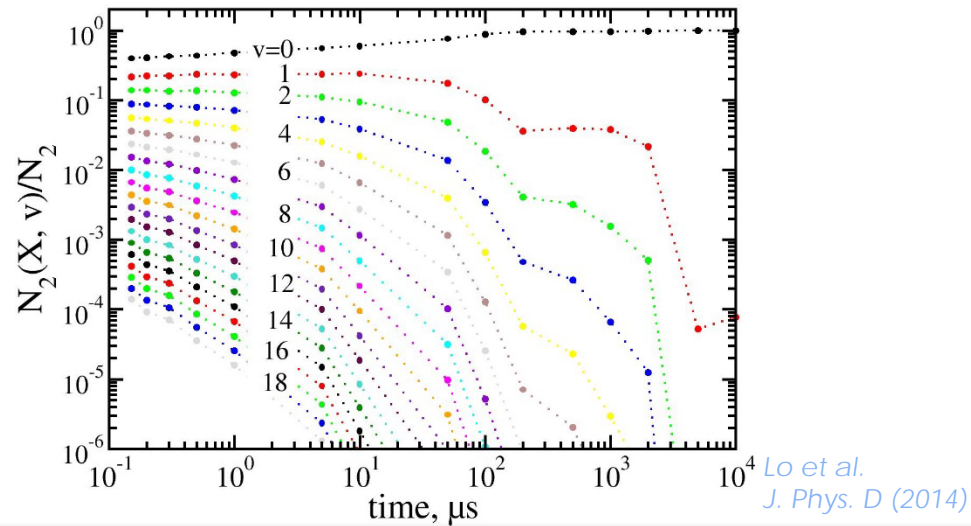
- 3 temperatures to describe the energetic state of $N_2(X)$
- T_{rot} : rotational temperature = T
 - Available from our experimental procedure despite the moderate spectral resolution
- $T_{v_{01}}$ relative to population of $v=0$ and $v=1$
 - Measurement of the vib. population up to $v=16$
- $T_{v_{1v}}$ Boltzmann distribution for $v \geq 1$

Two vibrational temperatures resulting for competition between V-T, V-V exchanges and e^- energy distribution function

N₂ RAMAN SPECTRA

Examples of N₂(X) spectra for ns-discharge in air
Averaged over 2000 shots



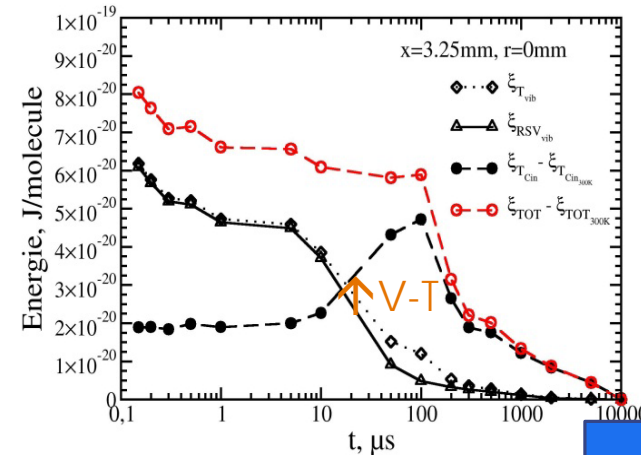
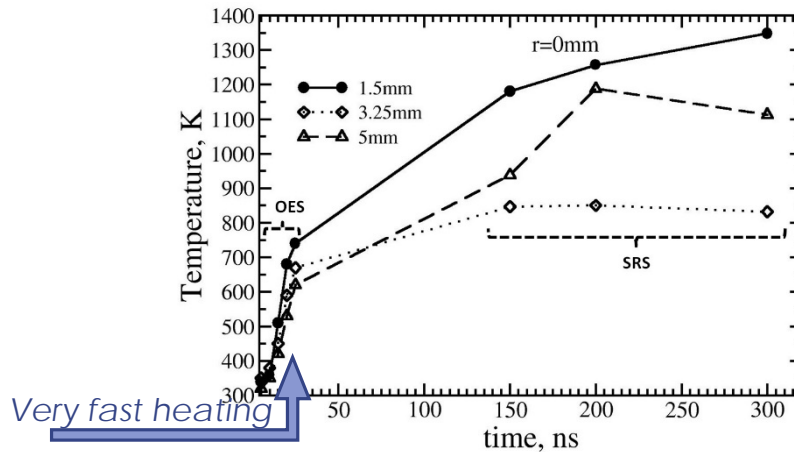


Spectrum modeling providing complete VDF
described by 3 temperatures

QUANTITATIVE CHARACTERIZATION OF ENERGY DEPOSITION

- SRS combined with emission spectroscopy

▶ very fast heating and time relaxation of the energy deposition
Most of energy transfer to gas



20 mJ deposited

- Quantification of energy on each mode of storage

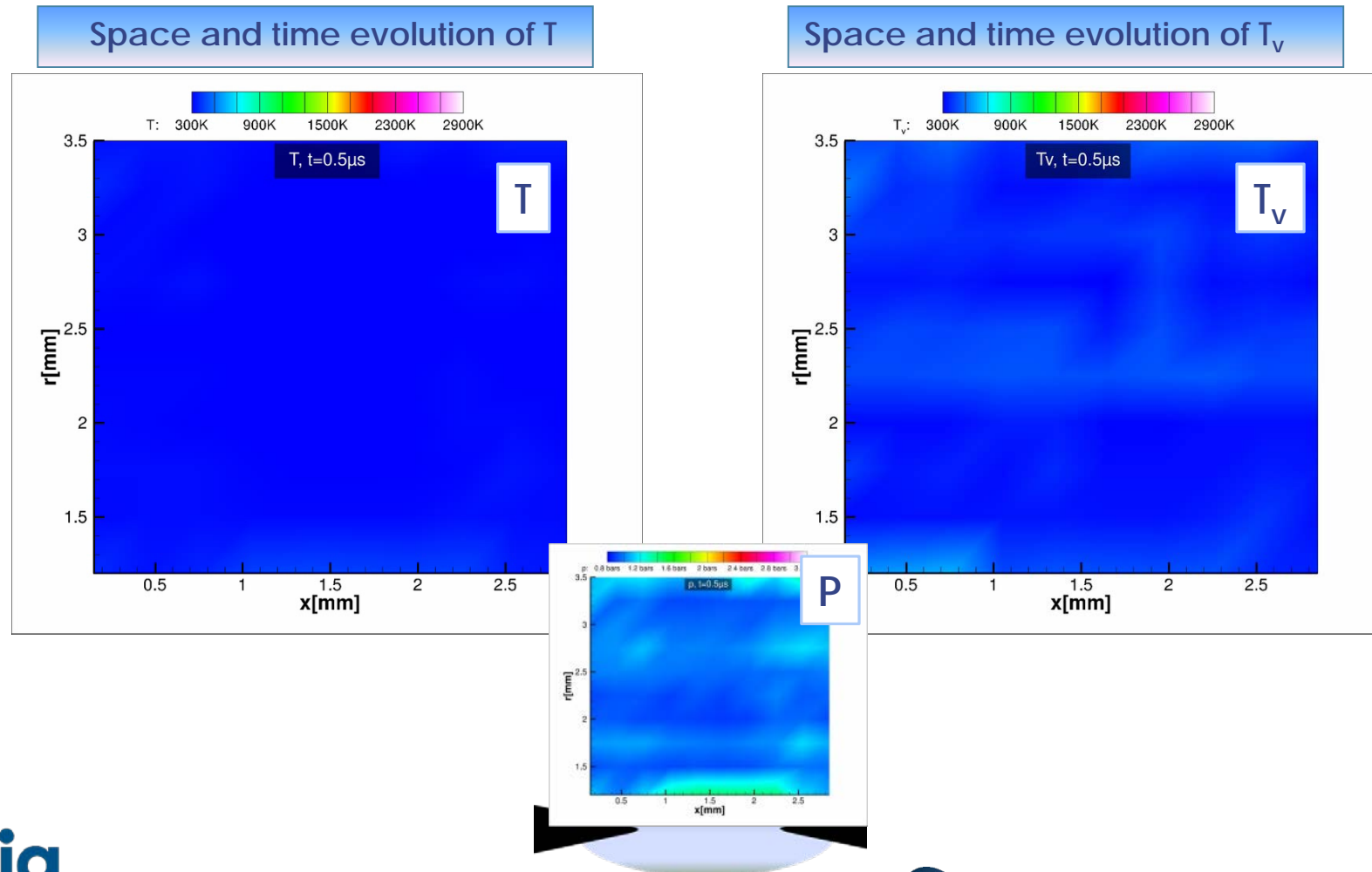
■ For N₂

- 150ns : 75% of energy in vibration
- 150ns -10μs: no transfer
- 10μs-100μs: V-T transfer
- After 200 μs : relaxation

Energy	N ₂	O ₂	Total
E_{Tot}^i (150 ns)	12.50 mJ @ 62.5%	5.00 mJ @ 25%	17.50 mJ @ 87.5%
$E_{T_{vib}}^i$ (150 ns)	9.50 mJ @ 47.5%	0.27 mJ @ 1.4%	9.77 mJ @ 48.9%
$E_{T_{Cin}}^i$ (150 ns)	3.00 mJ @ 15%	0.73 mJ @ 3.7%	3.73 mJ @ 18.7%
E_{Tot}^{dis} (150 ns)	—	4.00 mJ @ 20%	4.00 mJ @ 20%

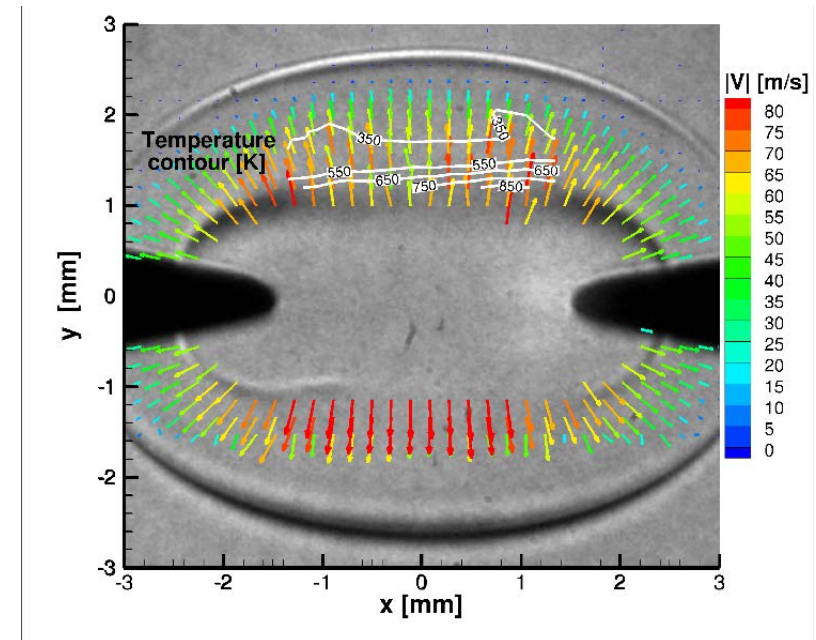
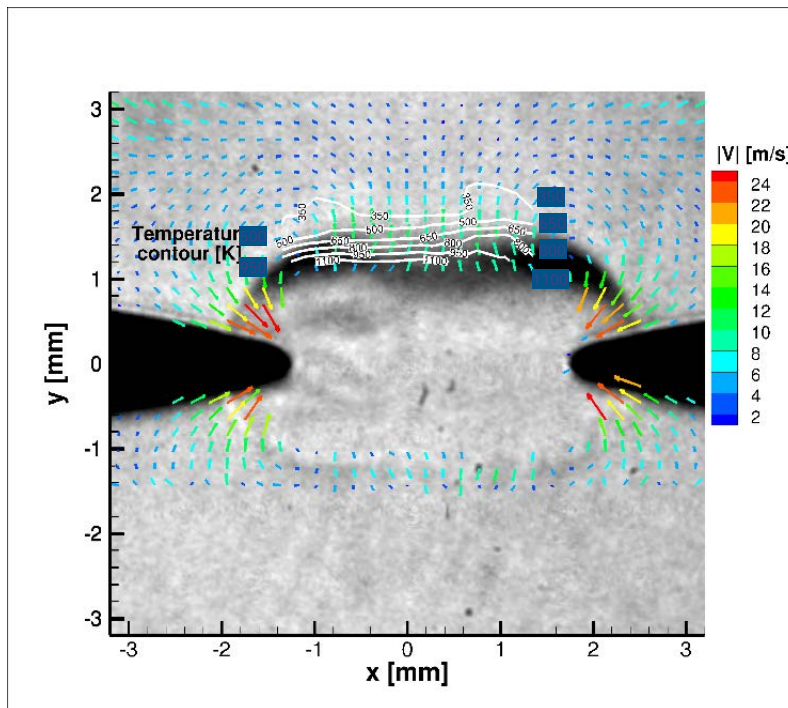
ANALYSE SPATIO-TEMPORELLE : EXEMPLE NS SPARK

T and T_v in the surrounding of the spark reconstruction par les profils 1D phasés



HYDRODYNAMIC EFFECTS

SRS combined with other diagnostics



EM2C's Schlieren images

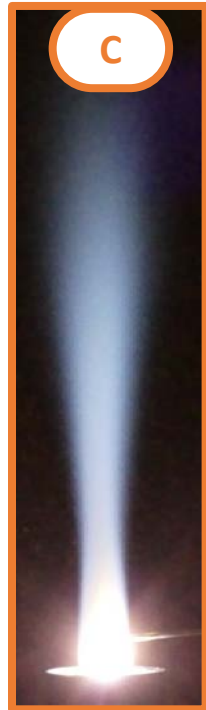
Combustion Assistée par plasma



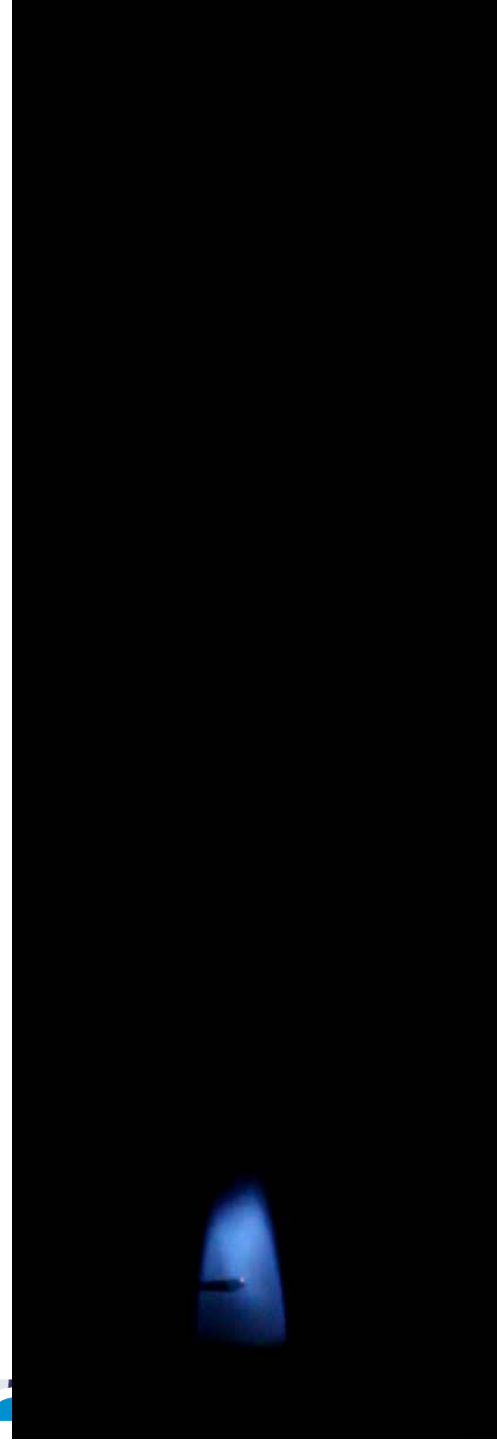
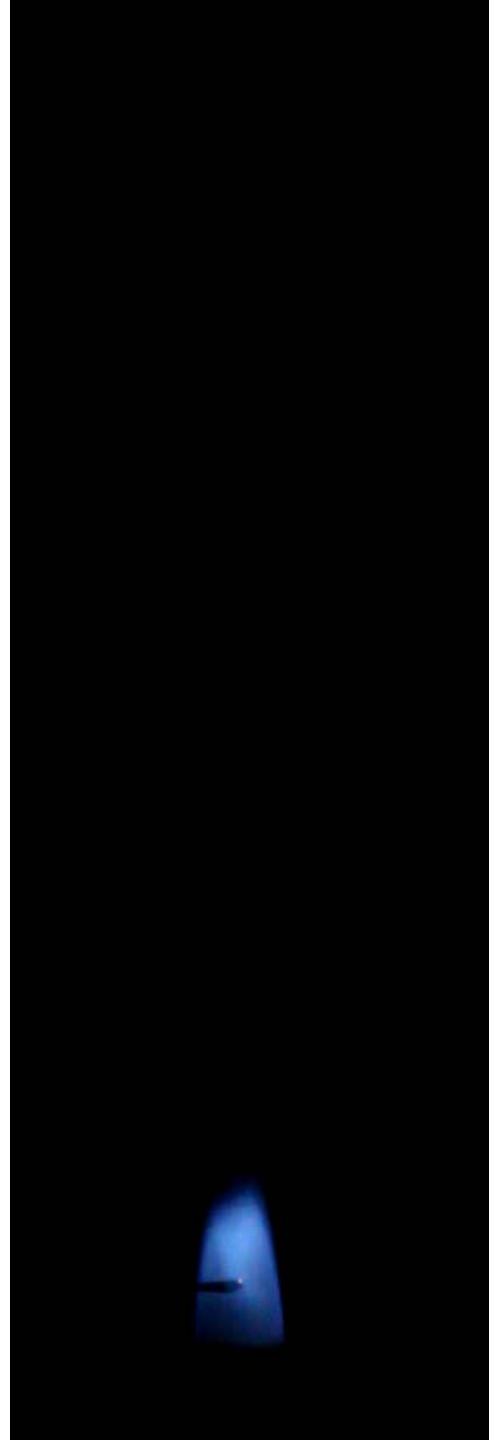
stable flame



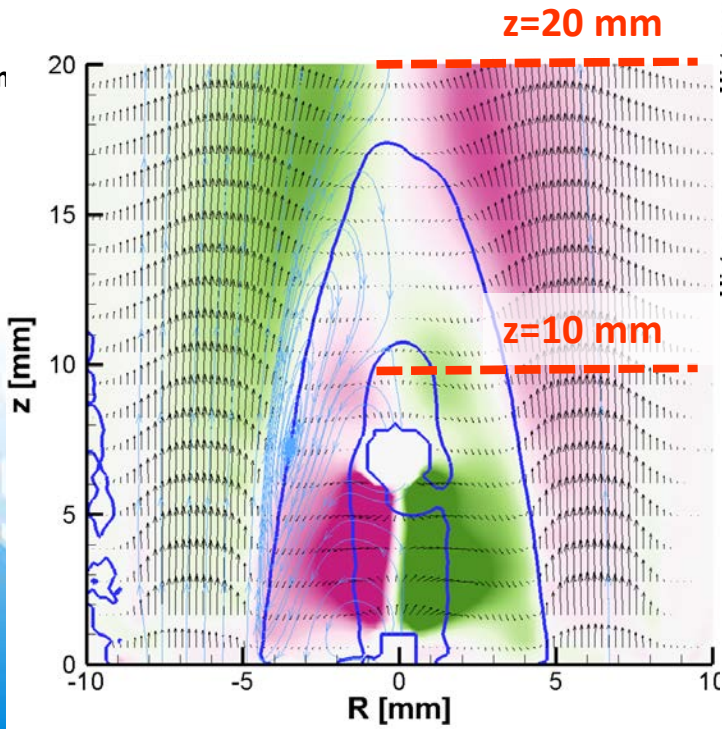
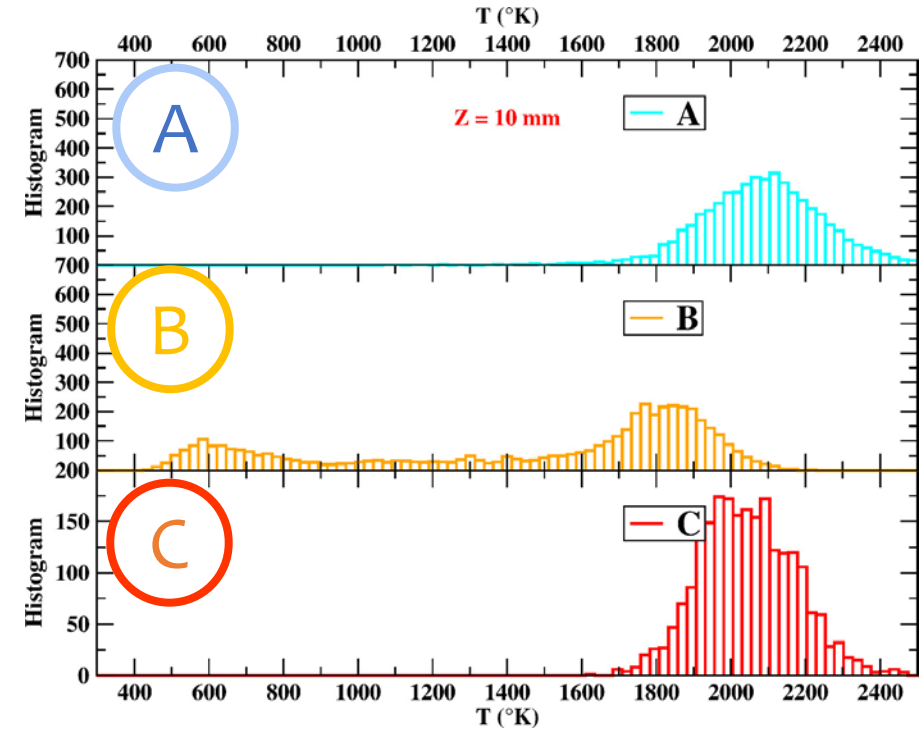
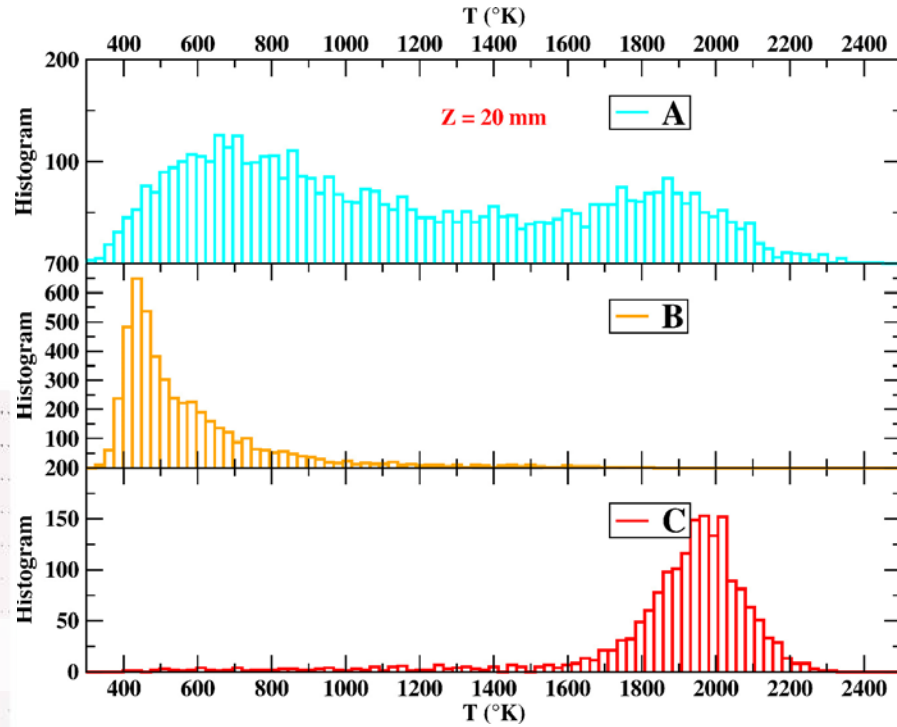
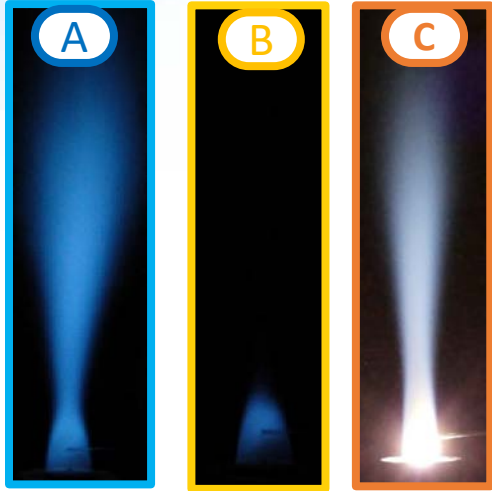
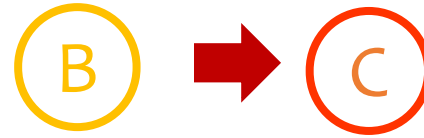
RZ flame



RZ flame
+NRP



EFFECT OF THE DISCHARGE



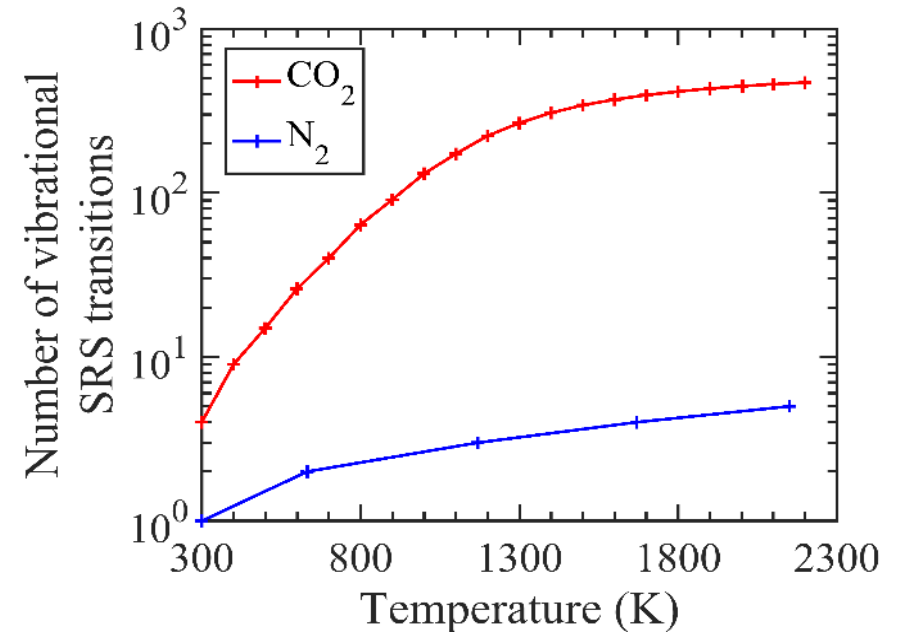
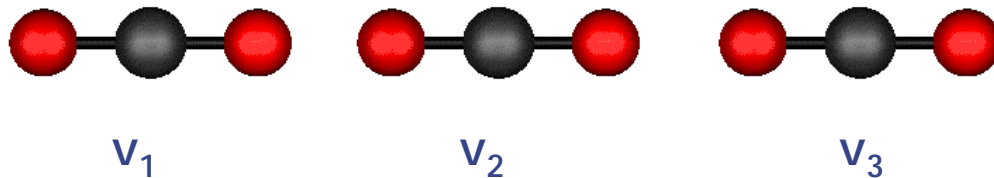
EXCITATION VIBRATIONNELLE DU CO2

difficulties?

Number of transitions

Confidence of spectroscopic data

Fermi resonance at high temperature



Hot temperature vibrational data base

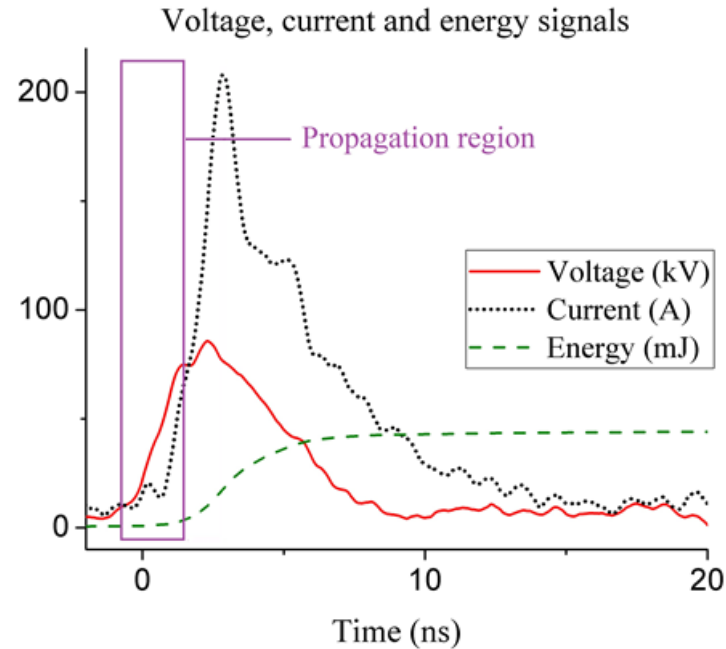
R. Lemus, et al(2014) J. Chem. Phys, 141

892 pure vibrational SRS transitions between 1100 and 1500 cm^{-1} , involving energy levels up to 15 000 cm^{-1}

VIBRATIONAL CO₂ SRS FOR NS-DISCHARGE ANALYSIS

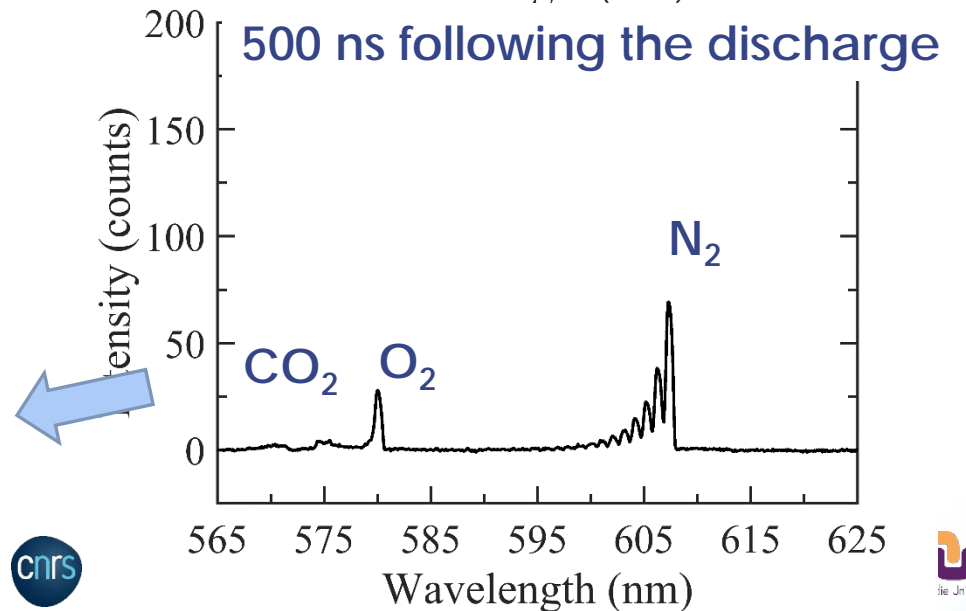
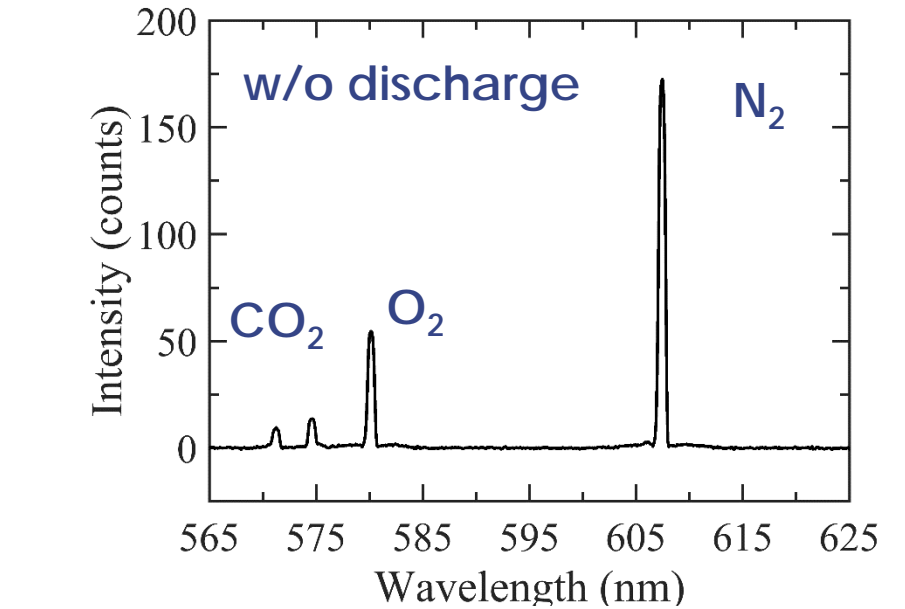
Air doped with 4% of CO₂

- Pin to plane discharge
- HV=85kV
- gap=18mm
- Total pulse duration: 10ns (FWHM=6ns)



Brisset, Tardiveau, GRC 2018

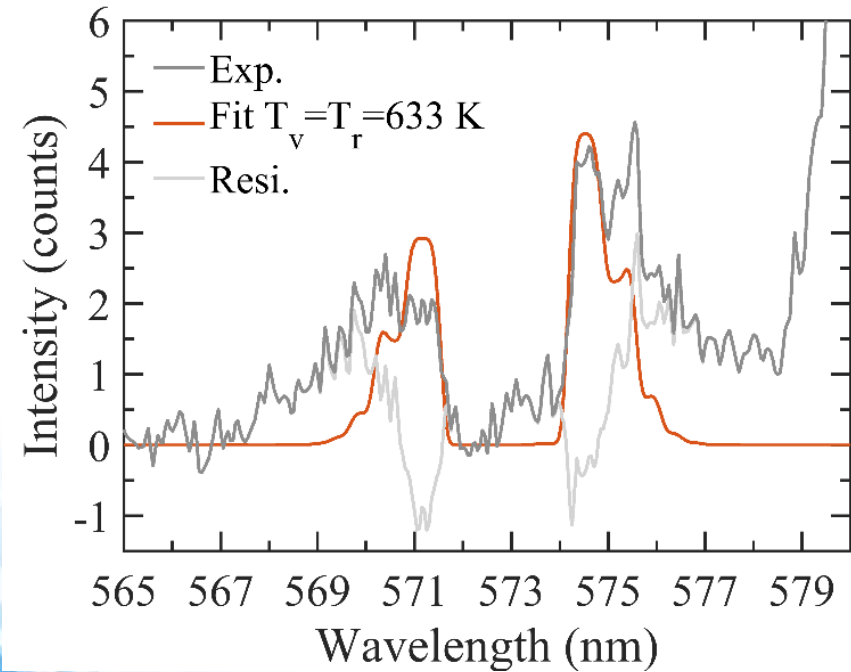
N₂ Tr= 671 K
 T_{v01}= 2583 K
 T_{v1v}= 4037 K



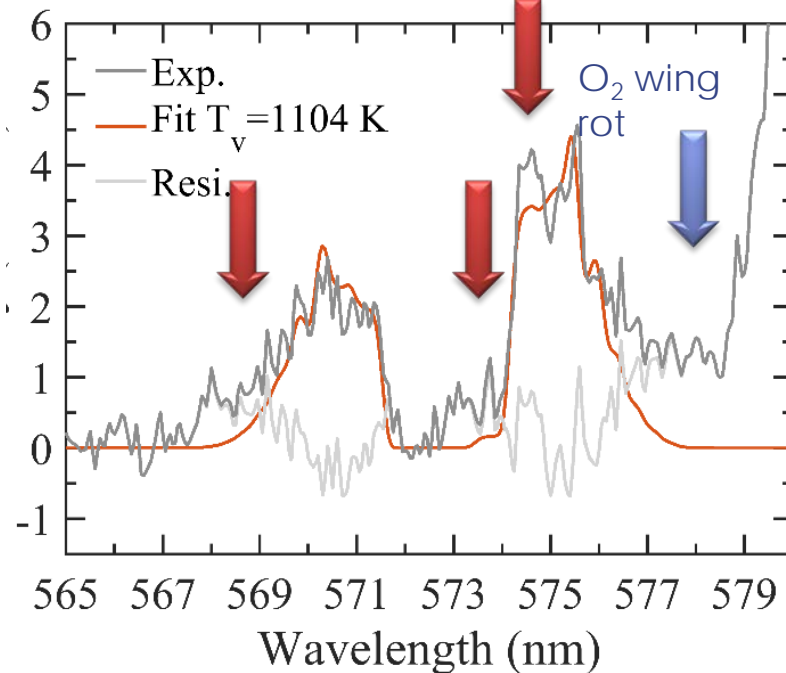
HORS ÉQUILIBRE DU SPRECTRE DE CO₂

$$f(E(1,2,1)) \propto \exp\left(-\frac{hc.E(1,2,0)}{kT_{12}}\right) * \exp\left(-\frac{hc.E(0,0,1)}{kT_3}\right)$$

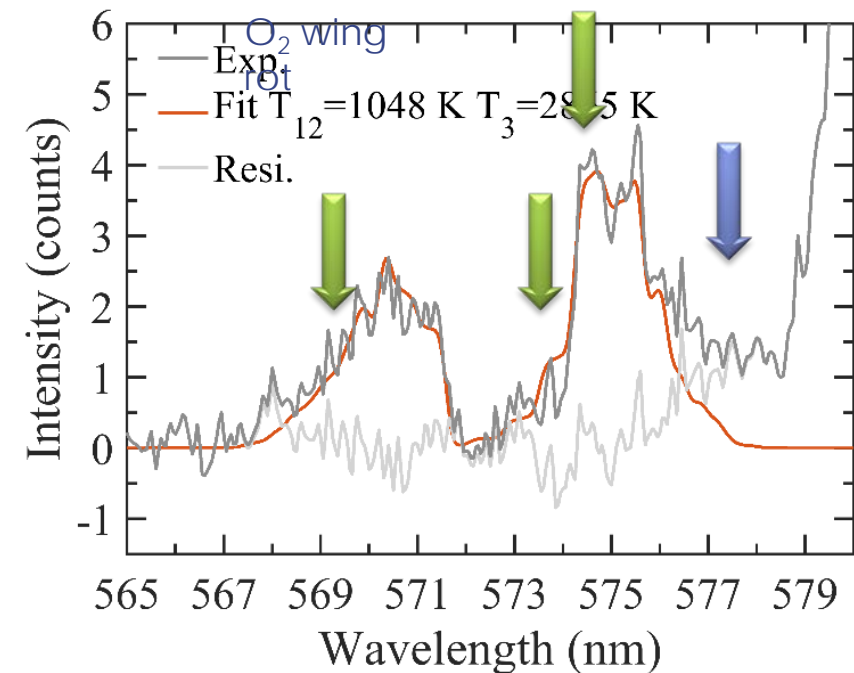
Simulation
 $T_v = T_r$



Simulation
 $T_v \neq T_r$

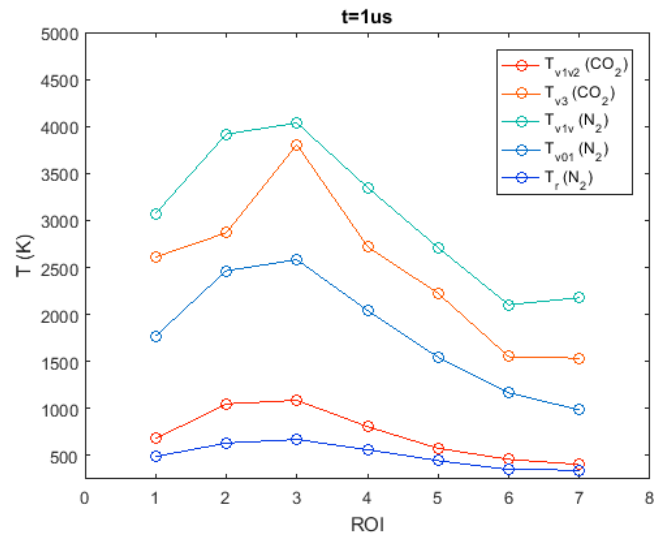
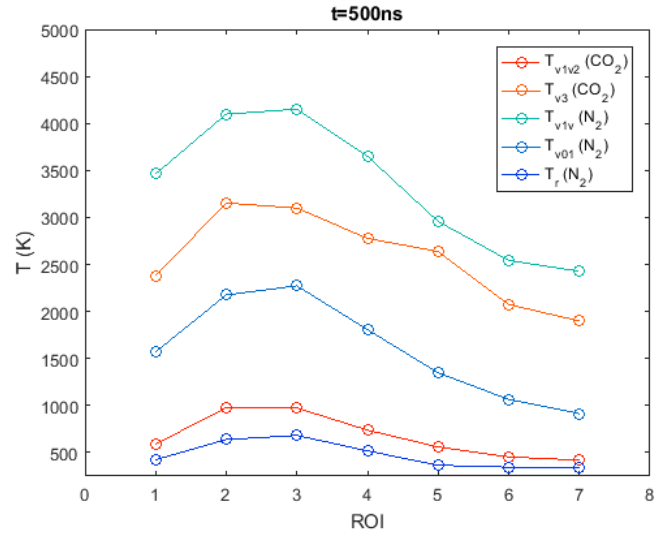


Simulation
 $T_v \neq T_r, T_{12} \neq T_3$

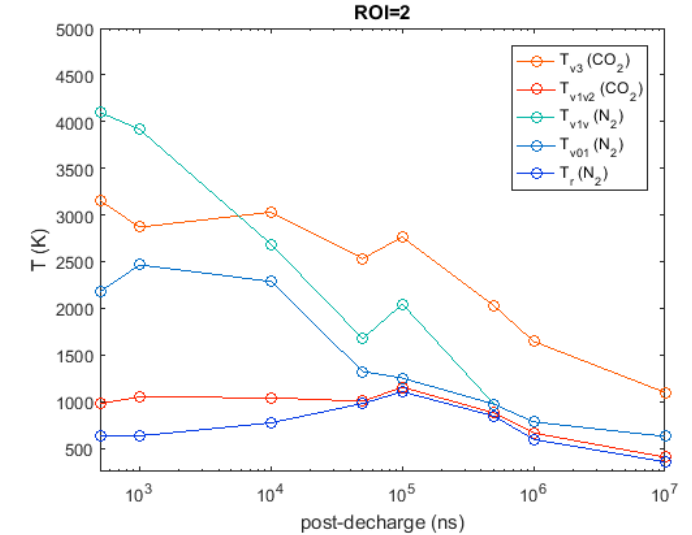
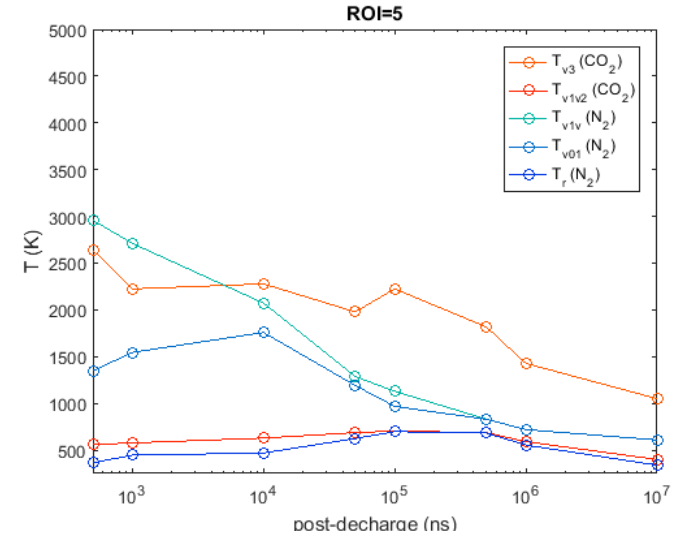


ANALYSE SPATIO-TEMPORELLE

Radial profiles



Axial time evolution



CONCLUSION

CONCLUSION

Diffusion Raman spontanée :
un diagnostic à revisiter

En combustion, en plasma, les 2 à la fois

**pour les plasmas hors équilibre : mesure
simultanée T et températures vibrationnelles**

mesure à Pa ou plus

mesure 1D : information spatio-
temporelle

moyenne de phase possible

**Intérêt pour les enjeux de la transition
énergétique**

Combustion turbulente : nouveaux
combustibles « défossilisés » en
combustion aérobie, en oxycombustion
diluée au CO₂, hydrogène

Combustion assistée par plasma

Valorisation du CO₂ : catalyse, catalyse
assistée par plasma

SIMULTANEOUS T AND MULTISPECIES CONCENTRATION

